

FScopp Database for FactSage 8.3

The FScopp Database is designed for thermodynamic and phase equilibrium calculations involving **Cu** alloys using the FactSage 8.0 Thermochemical Software Package (and later versions).

Cu Alloys	
Ag, Al, As, Au, Be, Bi, C, Ca, Cd, Co, Cr, Cu, Fe, Ga, Ge, H, Hg, In, Li, Mg, Mn, Mo, Na, Nb, Ni, O, P, Pb, Pd, Pt, S, Sb, Se, Si, Sn, Te, Ti, Tl, V, W, Zn, Zr	
Color codes	
Red : Cu Blue : Major alloying elements (full optimisations of binary systems with Cu and with several minor alloying elements, Cu-Xx-Yy ternary systems evaluated (good for the Cu-rich), several quaternary systems included); Green : Minor alloying elements (full optimisations of binary systems with Cu); Purple : Impurity elements (full optimisations of binary systems with Cu and several Cu-M-X systems); Black : Optimized for the Cu-Zz system and few Xx-Zz and Cu-Xx-Zz systems.	

Composition Ranges

The database is intended to allow calculations over all ranges of composition, although the assessed data are often most reliable for Cu-rich composition ranges. Please note that the FScopp database is a self-consistently evaluated database designed to be used independently of any other. Considerable caution must be exercised if it is used in conjunction with other FactSage alloy databases. However, calculations involving the gas phase can be performed with the FACTPS Database.

Temperature Ranges

The database is generally valid for the temperature range of approximately room temperature to 2200°C (please remember that kinetics effects can strongly affect phase relationships in alloys in general and in particular at low temperature), although for some alloys containing high melting point metals the data are reliable to still higher temperatures.

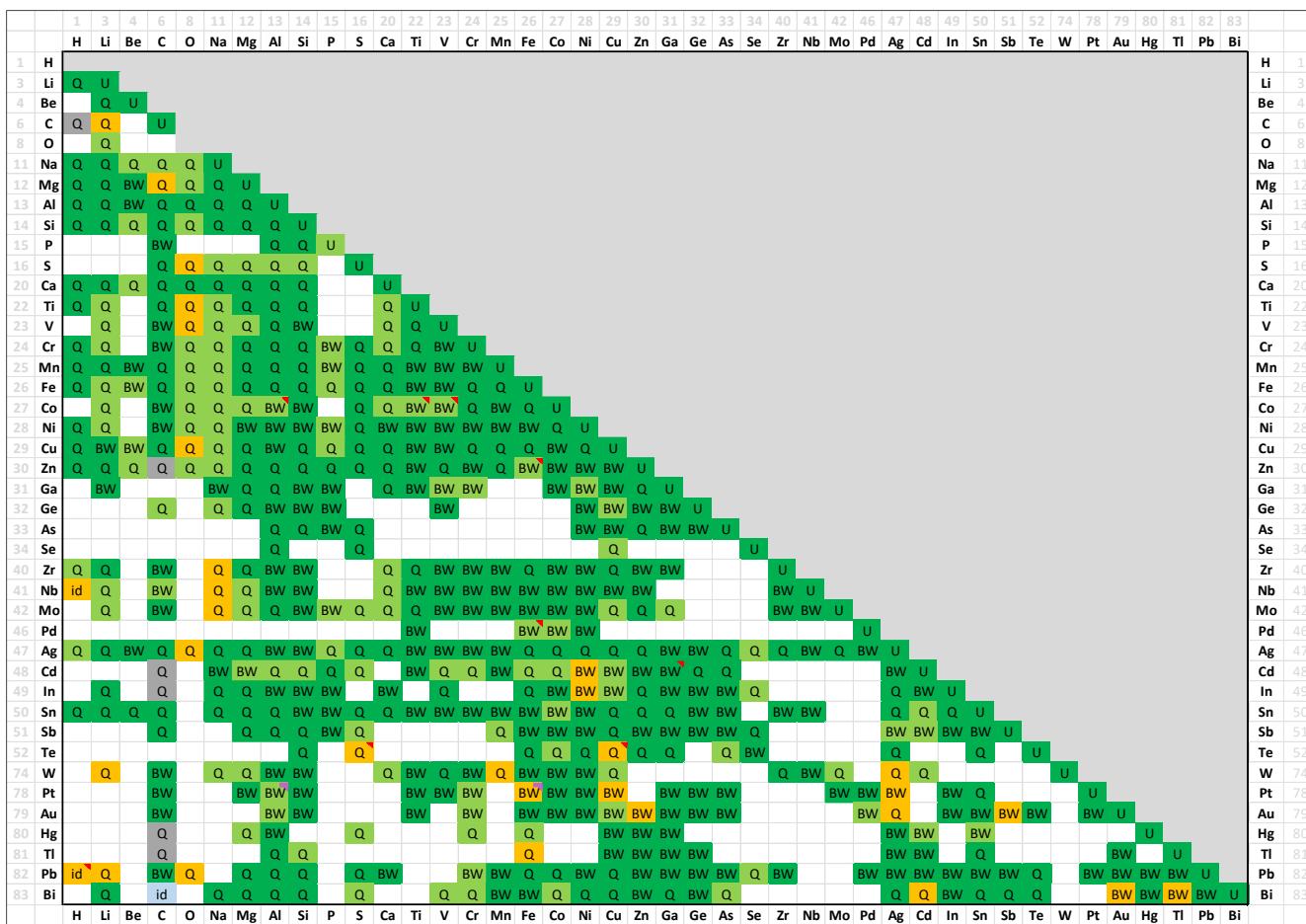
In **Table 4**, comments appearing in red can be found for some binary systems which have spurious stable phases at high temperatures, such as inverted liquid-liquid miscibility gaps, or the presence of an inverted solid solutions. The temperature above which the problem occurs is given in **Table 4** as an indication. These inverted stability regions can cause problems in the computation of liquidus

projection of ternary or higher order systems in the PHASE DIAGRAM Module and affect the computation of precipitation or formation targets in the EQUILIB Module.

Physical Properties

The density and viscosity of the liquid phase can be computed for most important elements present in the database.

The thermal conductivity of most non-metallic stoichiometric solids (ex.: SiC, Al₂O₃, etc.) can be computed with the FScopp 8.3 database. See FactSage (EQUILIB) documentation on physical properties, and the list of phases at the end of the present document.



id Ideal Bragg-Williams Treatment for the liquid, FCC-A1, BCC-A2, HCP-A3 and DHCP-A4 solid solutions

Q Modified Quasichemical Model (liquid solution); CEF for S.S.

BW Bragg-Williams Model (liquid solution); CEF for S.S.

Top quality

Good quality (maybe a missing phase, or a good estimation)

Rough estimate, probably missing phases

Noble gas system (use FACTPS Database)

Computed using pure substances from database with gaseous species from the FACTPS Database

Figure 1: Optimized binary systems in the FScopp 8.3 Database (see also Table 4)

In FScopp 8.3 a total of **603 binary systems** (515 in 8.2) have been evaluated, for most of them over the entire range of composition and for all stable phases. A matrix of assessed binary systems is given in Figure 1, for the major and minor alloying elements of Cu-alloys.

Several dozens of ternary systems have been assessed, and important quaternary systems have also been evaluated.

The FScopp 8.3 database contains **269 solution phases** (238 in 8.2) and **1077 pure compounds** (906 in 8.2) with **1442 stoichiometric phases** (1232 in 8.2) counting allotrophic forms. **Table 1** lists the important solutions for calculations involving Cu alloys. **Table 2** is a detailed list of all solutions. **Table 3** is a generic list of the pure compounds and their associated phases. **Table 4** gives the references for the binary data assessments.

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Use of the Database

The phase diagrams of all the binary systems listed above have been checked using FactSage 8.3.

Phase selection in the EQUILIB or PHASE DIAGRAM modules using FScopp is simple: simply follow these instructions:

- For pure solid compounds:
 - Right-click on “pure solids”
 - Then click “Select/Clear” > “Add all species from database” > “FScopp”
- For solutions:
 - In “Data Search | Options – search for product species”, select “minimum solution components = 2 cpts” (which is the default value);
 - Click on the “Select” button below the “Solution species” list box;
 - Then click “Add all phases from database” > “FScopp”;
 - Apply the recommendations related to the CBCC-A12, D82 and D85 solutions, as described in the warning text box in the following page.
- There is no need to select pure liquid phases (the FScopp-Liqu solution contains the liquid species). They may be selected as dormant (metastable, option “!”) for purposes of computing their chemical activity.
- Click “Use V & phys. property data” in the EQUILIB Module if you intend to have density, viscosity, thermal conductivity and surface tension to be calculated for phases, when available. We recommend not to click this option in the PHASE DIAGRAM Module.

There might be cases when a chemical system with many elements results in more than 150 possible solutions to be selected. In these cases, it is recommended to change the selection of some solutions where the “I-option” (“possible 2-phase immiscibility”) or the “J-option” (“possible 3-phase immiscibility”) has been automatically selected, to either “+” (for “I”) or “I-option” (for “J-option”). The solutions where this change must be done should then be selected at the user’s discretion based on his/her knowledge of which solutions might exhibit immiscibility in the composition range the user is interested in.

Within the solution database are hard coded default selections such as “+”, “I” or “J” (for single phase, 2-phase immiscibility and 3-phase immiscibility). However, when the system has more than 5 elements, a few specific solution phases will see their default selection changed to “!”, “!I” or “!J”, meaning they become “dormant” (metastable). If in the calculation(s) that will follow one of these phases has a calculated activity higher than 1.0, you should overwrite the “!” (dormant) selection.

CBCC-A12

It is recommended to select the CBCC-A12 solid solution with the “!I” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous potential end-members (1512) in that solution, which increases substantially the computation time. If after a calculation the activity of the CBCC-A12 solution is greater than 1.0, then redo the calculation with the standard “I” option.

D82 Prototype Cu₅Zn₈

Similarly it is recommended to select the D82 solid solution (Cu₅Zn₈, Al₈M₅ with M = Fe, Cr, V) with the “!I” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous potential end-members (8712) in that solution, which increases substantially the computation time. If after a calculation the activity of the D82 solution is greater than 1.0, then redo the calculation with the standard “I” option.

μ -phase Fe₇W₆ D85

Similarly it is recommended to select the D85 solid solution with the “!I” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous potential end-members (6561) in that solution, which increases substantially the computation time. If after a calculation the activity of the D88 solution is greater than 1.0, then redo the calculation with the standard “I” option.

Updates in 8.3 from 8.2

Here is a list of new assessments or new assessments integrated from publications in 8.3:

- 1) Al – Pt
- 2) As – Cd
- 3) As – Pt
- 4) As – Zn (replaced)
- 5) Au – Ga
- 6) Bi – Mn
- 7) C – Pt
- 8) Ca – Cu (replaced)
- 9) Ca – Ga
- 10) Ca – Ge
- 11) Cd – P
- 12) Co – Ga
- 13) Co – Pt (replaced)
- 14) Cr – Ga
- 15) Cu – Pt (replaced)
- 16) Fe – In (replaced)
- 17) Fe – Pt
- 18) Ga – Li
- 19) Ga – Mo
- 20) Ga – Ni
- 21) Ga – Pt
- 22) Ga – V
- 23) Ga – Zr
- 24) Ge – Na
- 25) Ge – Ni
- 26) Ge – Pt
- 27) Ge – V
- 28) H – Sn
- 29) In – Pt
- 30) In – Se
- 31) Li – Sb
- 32) Li – W
- 33) Mg – Pt
- 34) Mn – Sb
- 35) Mo – Pt
- 36) Ni – Pt

- 37) P – Sn
- 38) Pb – Pt
- 39) Pt – Si
- 40) Pt – Sn (replaced)
- 41) Pt – V

Many existing solutions were modified to introduce the new systems (*i.e.* Liquid metal, FCC-A1, BCC-A2, HCP-A3, etc.). Those are not necessarily listed below. Also, many existing solutions were modified to introduce new end-members:

- Diamond-A4 addition of In and V
- CBCC-A12 addition of Sb
- CUB-A13 addition of Sb
- A15 addition of Ga and Ge for Cr₃Ga, Mo₃Ga, V₃Ga, Cr₃Ge, Pt₃Ge, V₃Ge
- B2 AlPt, CoPt, CoTi, CoZr, MnNi
- B2b Au, Pt for PtTi, NiPt, AuTi
- B2e CoGa
- B8x (B81/B82) addition of Ge; updated CuSn, Au-Co, Ni₂Ge, Mn₂Sb
- B11 addition of Au for AuTi
- B20 addition of Pt, Ga, Ni, Mg
- B27 addition of Cr, Mo, Ni, Mn, V
- B31 addition of Pt Ga and Ge for PtGe, PtSi, NiGe
- B32 addition of Ga for LiGa
- B33 addition of Fe, Ge, Mo
- C1a addition of Ga, Mn, Pt for Ga₂Au, Ga₂Pt
- C2 addition of Ag, Pt, As, Se for PtSb₂, CoSe₂, CuSe₂, FeSe₂, NiSe₂
- C12 addition of Ge for CaGe₂
- C15 improvements to the Gibbs energy of metastable end-members
- C15a addition of Au and Bi
- C16 addition of Ga, Mg, Cr, Mo, V for Zr₂Ga, Mg₂Pt
- C23a addition of Ge and Pt for SnPt₂, SiPt₂, GeNi₂, GeCa₂, AlPt₂
- C36 addition of Mn
- C38 addition of Mn, Bi, S for Mn₂Sb
- C40 addition of Ge
- Ca addition of Mn
- Cb addition of Fe and Mo
- D03 improvements to the Gibbs energy of metastable end-members
- D011 addition of Ge, Mo, V
- D019 addition of Pt for MoPt₂;
- D022 improvements to the Gibbs energy of metastable end-members
addition of Pt for Pt₃V;
- D023 improvements to the Gibbs energy of metastable end-members
addition of Ga
- D024 addition of Au and Ga for Au₃Ga
- D13 addition of Ga

- D59 modified structure for Zn_3As_2 , Zn_3P_3 , Cd_3As_2 , Cd_3P_2
- D510 addition of Ni
- D513 addition of Ga and Pt for Al_3Pt_2 and Ga_3Pt_2
- D5a addition of Ga, Cr, Mo, Fe, V and Nb
- D85 many changes with Mo, W
- D83 addition of Ga, Ge, Cr, Fe, Mo
- L10 addition of Mn, Zr, Co, Pt, Ni
- L12 addition of Ga
- L12c addition of Ge and Pt for Ni_3Ge , Pt_3Al , Co_3Ti , Pt_3Co , Pt_3In
- L'2b addition of Pt for $SiPt_2$
- hP18 addition of Zr and Al for Zr_5Ga_4 , Zr_5Al_4
- mC18 addition of Nb
- mC22 addition of Ga
- mP24 re-organization of the structure for $ZnAs_2$, ZnP_2 , $CsAs_2$ and CdP_2
- mS20 addition of Pt
- mS44 addition of Pt and Ga
- oI6 addition of Pt for $MoPt_2$
- oP4 addition of Ti, Pt, Pd
- oP24 addition of P for SiP_2 and GeP_2
- tP20 addition of Ti

Moreover, the following new solid solution phases were created:

- 1) A3' Prototype La
addition of Au(Sn)
- 2) BCC-B2e Prototype CsCl
CoGa with vacancies
- 3) B19 Prototype Beta' AuCd
Au-Ti, (Mo,V,Ti)Pt
- 4) Bg Prototype-MoB
ZrGa
- 5) C1c Prototype CaF₂
Beta-Cd₃As₂ and betas-Zn₃As₂
- 6) D0c Prototype SiU₃
SiPt₃ is stable with Zr solubility
- 7) oD59 ordered ternary D59
- 8) D82a D82 prototype Cu₅Zn₈
V₆Ga₇
- 9) L11 L11 Prototype CuPt
CuPt
- 10) L12e L12 Prototype-AuCu₃
Pt₃Ga, Pt₃Ge, Ni₃Ga
- 11) h22P hP22 Prototype-Ti₆Sn₅
Ti₆Sn₅ and V₆Ga₅
- 12) h15R hR15 Prototype Al₂Li₃
(Al,Ga,In)₂Li₃
- 13) mS16 Prototype-Pt₃Ge
Pt₃(Ge,Si)
- 14) oF40 Prototype Al₃Zr₂
(Al,Ga)₃Zr₂
- 15) oS8 oS8 Prototype TII
GaCa, GeCa, NiZr
- 16) oS12 oS12 Prototype ZrGa₂
ZrGa₂
- 17) o16S oS16 Prototype Pt₅Ga₃
Ni₅Ga₃, Pt₅Ga₃
- 18) oS32 oS32 Prototype-Pu₃Pd₅
Zr₃Ga₅
- 19) t12I Prototype CdAs₂
CdAs₂

20) i160	tI160 Prototype Cd ₃ As ₂ (Cd,Zn) ₃ As ₂
21) tP16	Prototype Pt ₃ Ga Pt ₃ (Al,Ga)
22) p160	tP160 Prototype Zn ₃ As ₂ (Cd,Zn) ₃ As ₂
23) P	oP56 Frank-Kasper phase Ternary solution CrMoNi
24) P1	M3Ca ₂ (Cu,Mg) ₃ Ca ₂
25) P2	MCu ₃ (Ca,Mg)Cu ₃

Table 1: Important phases in Cu alloys present in the FScopp 8.3 Database

Common names	FScopp		Struktur-bericht	Pearson	Space Group (index)
	Soln-Nickname or CMPPD*	Solution Full name or compound stoichiometry			
Liquid metal	Liqu	Liquid	-	-	
Cu, α , FCC, (Cu), Ag, Co(lt), γ -Fe, Ni, Pb, Pt, Tl(ht)	A1	FCC-A1	A1	cF4	Fm-3m (225)
Cr, α -Fe, Li, Mo, Na, Nb, Ni, V, W, CuZn(ht), Cu ₃ Al(ht), Cu ₃ Be(ht), Cu ₃ In(ht), Cu ₃ Sn(ht)	A2	BCC-A2	A2	cI2	Im-3m (229)
Co, Tl(lt), Cu ₆ Si	A3	HCP-A3	A3	hP2	P6 ₃ /mmc (194)
Si also Ge, α -Sn	A4	Diamond-A4	A4	cF8	Fd-3m (227)
Bi	A7	Rhombehdral-A7	A7	hR2	R-3m (166)
C	-	C _(s)	A9	hP4	P6 ₃ /mmc (194)
Mn	A12	CBCC-A12	A12	cI58	I-4 ₃ m (217)
γ -Cu ₃₃ Si ₇ ('Cu ₅ Si')	-	Cu ₃₃ Si _{7(s)}	A13	cP20	P4 ₁ 3 ₂ (213)
β -NiZn, β' -CuZn, CuBe	B2	BCC-B2!BCC-A2	B2	cP2	Pm-3m (221)
ZnS-sphalerite	B3	B3 Prototype-ZnS	B3	cF8	F-43m (216)
ZnS-wurtzite	B4	B4 Prototype-ZnS	B4	hP4	P6 ₃ mc (186)
CuH	-	CuH _(s)	B4	hP4	P6 ₃ mc (186)
NiBi, NiAs, MnBi, AuSn, CoS, CrS, FeS, NiS, ε -FeSb, NiSb, (Fe,Ni)Te	B8x	B81/B82 NiAs/InNi2	B8 ₁	hP4	P6 ₃ /mmc (194)
Co ₂ Sb, InNi ₂ , (Fe,Mn,Ni,Ti) ₂ Sn, ε -FeSb, θ -SiNi ₂ , Sn(Fe,Ti) ₂			B8 ₂	hP6	
MnSi, (Co,Cr,Fe,Mn)Si	B20	B20 Prototype-FeSi	B20	cP8	P2 ₁ 3 (198)
Cu ₂ O	C3	C3 Prototype-Cu ₂ O	C3	cP6	Pn-3m (224)
Cu(Ti,Zr) ₂	C11b	C11b Prototype-MoSi2	C11 _b	tI6	I4/mmm (139)
Cu ₂ Mg	C14	C14 Prototype-MgZn2	C14	hP12	P6 ₃ /mmc (194)
CdCu ₂	-	CdCu _{2(s)}	C14	hP12	P6 ₃ /mmc (194)
Cu ₂ Mg, Cu ₂ Zr	C15	C15 Prototype-MgCu2	C15	cF24	Fd-3m (227)

Cu₉Zr₂ ('Cu₅Zr')	-	Cu ₉ Zr _{2(s)}	C15 _b	cF24	F-43m (216)
Cu ₂ Te(s4), η-Cu ₂ Sb	C38	C38 Prototype-Cu ₂ Sb	C38	tP6	P4/nmm (129)
Cu ₂ Te(s5)	Ch	Ch Prototype-Cu ₂ Te	C _h	hP6	P6/mmm (191)
Cu ₃ P	D021	D021 Prototype-Cu ₃ P	D0 ₂₁	hP24	P6 ₃ cm (185)
Cu ₃ Sn(ht), Cu ₃ Sb, Mn ₃ Si	D03	D03 Prototype-BiF ₃	D0 ₃	cF16	Fm-3m (225)
Fe₃Si	D03a	D03 Prototype-BiF ₃	D0 ₃	cF16	Fm-3m (225)
CaCu ₅	D2d	D2d Prototype-CaCu ₅	D2 _d	hP6	P6/mmm (191)
Mn ₉ Si ₂	-	Mn ₉ Si _{2(s)}	D5 ₉	tP10	P4/mbm (127)
Cu₂Al(lt)	D83	D83 Prototype-Cu ₉ Al ₄	D8 ₃	cP52	P-4 ₃ m (215)
Mn ₅ Si ₃	D88	D88 Prototype-Mn ₅ Si ₃	D8 ₈	hP16	P6 ₃ /mmc (193)
δ-Cu ₇ In ₃	aP40	aP40 Prototype-Cu ₇ In ₃	-	aP40	P-1 (2)
Cu₂S(ht)-digenite, Cu₂Se(ht)- berzelianite, (Cu_{2-x}Fe_x)S-bornite	cF44	cF44 Prototype-Cu ₂ Se	-	cF44	Fm-3m (225)
Cu₂S(mt)-high-chalcocite	hP16	hP16 (high-chalcocite)	-	hP16	P6 ₃ /mmc (194)
Cu₂S(lt)-chalcocite	Chal	Chalcocite	-	mP144	P2 ₁ /c (14)
β'-Cu₄Ti	oP20	oP20 Prototype-ZrAu ₄	-	oP20	Pnma (62)
Cu ₂ Te _(s6)	-	Cu ₂ Te _(s6)	-	cF12	Fd-3m (227)
ε-AsCu ₈	-	AsCu _{8(s)}	-	h*	P6 ₃ /mmc (194)
Cu ₃ As-beta-Domeykite	-	Cu ₃ As _(s)	-	hP24	P-3c1 (165)
Cu ₅₁ Zr ₁₄	-	Cu ₅₁ Zr _{14(s)}	-	hP68	P6/m (175)
Hg ₆ Cu ₇	-	Hg ₆ Cu _{7(s)}	-	hR*	R3m
Cu₂Se(lt)	-	Cu ₂ Se _(s)	-	mS144	C2/c (15)
Cu₃Sn	-	Cu ₃ Sn _(s)	-	oS80	Cmcm (63)
Cu ₂ Te _(s1) (rt)	-	Cu ₂ Te _(s1)	-	o*	?
Cu ₂ Te _(s2)	-	Cu ₂ Te _(s2)	-	o*	?
Cu ₂ Te _(s3)	-	Cu ₂ Te _(s3)	-	?	?
δ-Cu ₄ Sb	-	Cu ₄ Sb _(s)	-	?	?
γ-Cu ₆ Sb	-	Cu ₆ Sb _(s)	-	?	?

Table 2: Solution names with their major components (not including minor solutes)

■ Important solutions in Cu alloys

Common names	FScopp		Major phase components
	Nickname	Full name	
Liquid metal	Liqu	Liquid	most elements
Cu, (Cu), FCC	A1	FCC-A1	Ag, Al, Cu, γ -Fe _{austenite} , Ni, Pb, ... with C, H, O interstitials (\rightarrow B1)
BCC	A2	BCC-A2	α -Fe _{ferrite} , Cr, Li, Na, β -Ti, β -Zr, Mo, Nb, V, W... ... with C, H, O interstitials
Co, HCP, α -Ti	A3	HCP-A3	Co, Mg, α -Ti, α -Zr ... with C, H, O interstitials
Au(Sn)	A3'	DHCP-A3'	Au-rich solution with Sn
Zn	A3''	HCP-Zn	Zn
Si	A4	Diamond-A4	C _{diamond} , Ge, Si, α -Sn
Sn	A5	BCT-A5	β -Sn
In	A6	TET-A6	α -In
InSn	A6o	TET-A6o	β -InSn
As, Bi, Sb	A7	RHOM-A7	α -As, Bi, Sb
γ -Se, Te	A8	A8	γ -Se, Te
Hg	A10	RHOM-A10	α -Hg
Ga	A11	ORTH-A11	α -Ga
α -Mn, χ - (Mo,W) ₅ Fe ₁₂ Cr ₁₂ , (γ - or β -) Al ₁₂ Mg ₁₇	A12	CBCC-A12	α -Mn, β / γ -Al ₁₂ Mg ₁₇ , χ -FeCr(Mo,W)
β -Mn	A13	CUB-A13	β -Mn, Ag ₃ Al
Cr ₃ Si	A15	CUB-A15	Cr ₃ Si, Nb ₃ Al, Nb ₃ Sn, V ₃ Co, V ₃ Ni, V ₃ Si, V ₃ Sn, Mo ₃ Al
α -S	A16	A16	S
γ -InSn	Af	HEX-Af	γ -InSn
β -S	Ak	Ak	β -S
Se	Al	Al	Se
P(red)	RedP	Red-Phosphorus	P(red)
CaS, MgS, MnS, PbS	B1	FCC-B1	CaS, MgS, MnS, PbS, PbSe, PbTe, SnAs, SnS, SnTe
β -NiZn, β' -CuZn	B2	BCC_B2!BCC_A2	AgLi, AgMg, CaIn, β -NiZn, β' -CuZn with order/disorder transf.
TiNi, TiAu, TiPt	B2_b	BCC-B2b	non-stoichiometric TiNi
AlNi	B2_c	BCC-B2c	non-stoichiometric AlNi with vacancies
AlMo	B2_d	BCC-B2d	non-stoichiometric AlMo
CoGa	B2_e	BCC-B2e	non-stoichiometric CoGa
ZnS, ZnTe, AlP, β -SiC-3C	B3	B3 (sphalerite)	AlAs, AlP, AlSb, GaAs, GaP, GaSb, InAs, InP, InSb, β -SiC-3C

ZnS, α -SiC-2H	B4	B4 (wurtzite)	ZnS, α -SiC-2H
(Co,Fe,Cr,Ni)S, B8 ₁	B8x	B81/B82 NiAs/InNi ₂	B8₁ : ε -AlCu, NiBi, NiAs, MnBi, AuSn, CoS, CrS, FeS, NiS, ε -FeSb, NiSb, (Fe,Ni)Te
			B8₂ : Co ₂ Sb, β -BiMn, InNi ₂ , (Fe,Mn,Ni,Ti) ₂ Sn, ε -FeSb, θ -SiNi ₂ , GaTi ₂ , Sn(Fe,Ti) ₂
HgS	B9	B9 Prototype-HgS	HgS
γ -CuTi	B11	B11 Prototype-CuTi	γ -CuTi, Ag(Ti,Zr), β 1-NiZn, CdTi
NiS millerite	B13	B13 Prototype-NiS	NiS
SnS, α -SnSe	B16	B16 Prototype-GeS	SnS, α -SnSe
CuS covellite	B18	B18 Prototype-CuS	CuS
(Au,Pt)Ti	B19	B19 Prototype-Beta'AuCd	AuTi, PtTi, MoPt
FeSi	B20	B20 Prototype-FeSi	(Fe,Co,Cr,Mn)Si monosilicide
TiSi, ZrSi	B27	B27 Prototype-FeB	TiSi, ZrSi
FeP, MnP, CoP, NiSi	B31	B31 Prototype-FeAs	(Co,Fe,Mn)P, NiSi
ZnLi, InLi, AlLi	B32	B32 Prototype-NaTl	AlLi, InLi, InNa, ZnLi
CaSi, CaSn	B33	B33 Prototype-CrB	CaSi, CaSn
CoSn, Niln, FeSn	B35	B35 Prototype-CoSn	CoSn, FeSn, Niln
η -AgZn	Bb	Bb Prototype-AgZn	η -AgZn (also called β')
ω -CdSb, β -ZnSb	Be	Be Prototype-CdSb	ω -CdSb, β -ZnSb
ZrGa	Bg	Bg Prototype-MoB	ZrGa
γ -MoC, δ -WC	Bh	Bh Prototype-MoC	γ -MoC, δ -WC
TiH ₂	C1	C1 Prototype-CaF ₂	TiH ₂ , δ -ZrH ₂
Mg ₂ Si	C1a	aC1 Prototype -CaF ₂	Mg ₂ Si, Mg ₂ Sn, Mg ₂ Ge, Mg ₂ Pb, Al ₂ Au, Si ₂ Co, Si ₂ Ni
Ni ₂ S- β 1	C1b	C1b Prototype-MgAgAs	Ni ₂ S- β 1
(Cd,Zn) ₃ As ₂ (ht)	C1c	C1c Prototype-CaF ₂	(Cd,Zn) ₃ As ₂ (ht)
(Co,Fe,Mn,Ni)S ₂ , (Co,Cu,Mn,Ni)Se ₂ , AuSb ₂	C2	C2 Prototype-FeS ₂	(Co,Fe,Mn,Ni)S ₂ , AuSb ₂ , (Co,Cu,Mn,Ni)Se ₂ ,
Cu ₂ O	C3	C3 Prototype-Cu ₂ O	(Cu,Ag) ₂ O
MTe ₂ , SnS ₂	C6	C6 Prototype-CdI ₂	(Fe,Ni,Co)Te _{2-x} , SiTe ₂ , SnS ₂ , ZrS ₂ and SnSe ₂
MoS ₂	C7	C7 Prototype-MoS ₂	MoS ₂ , MoSe ₂ , MoTe ₂
CuTi ₂	C11b	C11b Prototype-MoSi ₂	MoSi ₂ , AgTi ₂ , AgZr ₂ , AlCr ₂ , CuTi ₂ , CuZr ₂ , ZnTi ₂ , TiAu ₂ , TiPd ₂
CaSi ₂	C12	C12 Prototype-CaSi ₂	CaSi ₂
MgZn₂, Laves-C14, Hexagonal Laves	C14	C14 Prototype-MgZn ₂ (Laves)	Al ₂ Zr, Co ₂ Mg, Cr ₂ Nb, Cr ₂ Ti, Cr ₂ Zr, Cu ₂ Mg, Fe ₂ Mo, Fe ₂ Nb, Fe ₂ Ti, Fe ₂ W, Li ₂ Ca, Mg ₂ Ca, Mn ₂ Nb, Mn ₂ Ti, Mn ₂ Zr, Ti ₂ Al, Zn ₂ Mg, Zn ₂ Ti

MgCu₂, Laves-C15, Cubic Laves	C15	C15 Prototype-MgCu ₂ (Laves)	Al ₂ Ca, Co ₂ Ti, Co ₂ Zr, Co ₂ Nb, Cr ₂ Ti, Cr ₂ Zr, Cr ₂ Nb, Cu ₂ Mg, Cu ₂ Zr, Fe ₂ Ti, Fe ₂ Zr, Ni ₂ Ca, V ₂ Zr, Zn ₂ Zr
Laves-Ag ₂ Na	C15a	C15a Prototype-MgCu ₂ (Laves)	Ag ₂ Na
Laves-Ni ₅ Zr	C15b	C15b Prototype-MgSnCu ₄ (Laves)	Ni ₅ Zr
θ-Al ₂ Cu	C16	C16 Prototype-Al ₂ Cu	θ-Al ₂ Cu, Pb ₂ Au, Sn ₂ (Co,Fe,Mn), Zr ₂ Si, (Ti,Zr) ₂ Ni, Zr ₂ (Co,Fe,Ni,Si)
CuSe ₂ , (Co,Fe,Ni)(As,S,Sb) ₂	C18	C18 Prototype-FeS ₂	CuSe ₂ , (Co,Fe,Ni)(As,S,Sb,Te) ₂
Fe ₂ P	C22	C22 Prototype-Fe ₂ P	Cu ₂ P, Fe ₂ P, Mn ₂ P, Ni ₂ P, FeNiP
Co ₂ Si	C23a	C23 Prototype-Co ₂ Si	Ca ₂ Sn, Sr ₂ Sn, Ca ₂ Si, Sr ₂ Si, θ-Ni ₂ Si, Ca ₂ Cu, Co ₂ Si, PbCa ₂
Bi ₂ Te ₃	C33	C33 Prototype-Bi ₂ Te ₃	Bi ₂ Te ₃ , Bi ₂ Se ₃ , Sb ₂ Te ₃
AuTe ₂	C34	C34 Prototype-AuTe ₂	AuTe ₂
Laves-MgNi ₂	C36	C36 Prototype-MgNi ₂ (Laves)	MgNi ₂ , Cr ₂ (Ti,Zr), Cu ₂ Mg, ternary phases in Mg-Al-Ca, Co ₂ Ti
η-Cu ₂ Sb	C38	C38 Prototype-Cu ₂ Sb	η-Cu ₂ Sb, Cu ₂ As, Cu ₂ Te, Fe ₂ Te
CrSi ₂	C40	C40 Prototype-CrSi ₂	CrSi ₂ , NbSi ₂ , VSi ₂ , WAl ₂ , WSi ₂
CaZn ₂	C42	C42 Prototype-CeCu ₂	CaZn ₂
Si ₂ Zr	C49	C49 Prototype-Si ₂ Zr	Si ₂ Zr, Si ₂ Ti
TiSi ₂	C54	C54 Prototype-TiSi ₂	TiSi ₂ , ZrSn ₂
NiMg ₂ C _a	Ca	Ca Prototype-NiMg ₂	NiMg ₂
CuMg ₂	Cb	Cb Prototype-CuMg ₂	CuMg ₂ , NbSn ₂ , VSn ₂
Cu ₂ Te	Ch	Ch Prototype-Cu ₂ Te	Cu ₂ Te
SiPt ₃	D0c	D0c Prototype-SiU ₃	SiPt ₃
M ₃ P (M=Co,Cr,Fe, Mn,Ni,V)	D0e	D0e Prototype-Ni3P	
Kieftite Sb ₃ Co	D02	D02 Prototype-As ₃ Co	Sb ₃ Co
β-Ni ₃ Sn, γ-Cu ₃ Sn D0 ₃	D03	D03 Prototype-BiF ₃	γ-Cu ₃ Sn, Sn ₃ Cu, β-Ni ₃ Sn, Ni ₃ Sb, Cu ₃ Sb, Mn ₃ Si, Li ₃ Bi
Fe₃Si	D03a	D03 Prototype-BiF ₃	Fe ₃ Si
Li ₂ MgSn	D03b	D03b Prototype-BiF ₃	Li ₂ MgSn
θ-Fe ₃ C, Al ₃ Ni	D011	D011 Prototype-Fe ₃ C	θ-Fe ₃ C, Mn ₃ C, Al ₃ Ni, Ni ₃ Si
Na ₃ As, Na ₃ P	D018	D018 Prototype-Na ₃ As	Na ₃ As, Na ₃ Bi, Na ₃ P, Na ₃ Sb
α ₂ -Ti ₃ Al D0 ₁₉	D019	D019 Prototype-Ni3Sn	Nb ₃ Al, Ti ₃ Al, Ti ₃ Ga, Zr ₃ Al, Ti ₃ Sn, Ni ₃ Zr, Ni ₃ In, Ni ₃ Sn(lt), Co ₃ W, Mo ₃ Al, Mn ₃ Sn
Cu ₃ P	D021	D021 Prototype-Cu ₃ P	Cu ₃ P
Al ₃ X D0 ₂₂	D022	D022 Prototype-TiAl ₃	Al ₃ Cr, Al ₃ Mo, Al ₃ Nb, ε-Al ₃ Ti, Al ₃ V, Al ₃ Zr, Ga ₃ Ti, Ni ₃ V
Al ₃ Zr D0 ₂₃	D023	D023 Prototype-ZrAl ₃	Al ₃ Zr
Ni ₃ Ti D0 ₂₄	D024	D024 Prototype-Ni ₃ Ti	Ni ₃ Ti
Ag ₃ Sn, Cu ₃ Sb D0 _a	D0a	D0a Prototype-Cu ₃ Ti	Ag ₃ Sn, Cu ₃ Sb, γ-Ni ₃ Mo, Ni ₃ Nb, Ni ₃ Sb

$\text{Cr}_7\text{C}_3(\text{hT})$, Mn_7C_3 D10 ₁	D101	D101 Prototype- Mn_7C_3	Cr_7C_3 , Mn_7C_3
$\text{Cr}_7\text{C}_3(\text{IT})$	D102	D102 Prototype- Th_7Fe_3	Cr_7C_3
$\text{Al}_4(\text{Ca}, \text{Sr}, \text{Ba})$	D13	D13 Prototype- Al_4Ba	Al_4Ca
Al_2CaZn_2	D13a	D13a Al_2CaZn_2	Al_2CaZn_2 , Al_2REZn_2 (with $\text{RE} = \text{La}, \text{Ce}, \text{Pr}, \text{Nd}$)
TiAu_4 , MoNi_4 , WNi_4	D1a	D1a Prototype- MoNi_4	TiAu_4 , MoNi_4 , WNi_4
$\text{Mg}_2\text{Zn}_{11}$, $\text{Mg}_2\text{Cu}_6\text{Al}_5$	D22	D22 Prototype- $\text{Mg}_2\text{Zn}_{11}$	$\text{Mg}_2\text{Zn}_{11}$, $\text{Mg}_2\text{Cu}_6\text{Al}_5$
NaZn_{13}	D23	D23 Prototype- NaZn_{13}	NaZn_{13} , CaZn_{13}
$\text{Ca}(\text{Cu}, \text{Ni}, \text{Zn})_5$	D2d	D2d Prototype- CaCu_5	CaZn_5 , CaCu_5 , CaNi_5
Al_6Mn , $\text{Al}_6(\text{Fe}, \text{Mn})$	D2h	D2h Prototype- Al_6Mn	Al_6Mn (Al_6Fe) _{metastable}
$\alpha\text{-Mg}_3(\text{Bi}, \text{Sb})_2$	D52	D52 Prototype- La_2O_3	$\text{Mg}_3(\text{Bi}, \text{Sb})_2$
$\beta\text{-Mg}_3(\text{Bi}, \text{Sb})_2$	D53	D53 Prototype- Mn_2O_3	$\text{Mg}_3(\text{Bi}, \text{Sb})_2$
$(\text{Bi}, \text{Sn})_2\text{S}_3$	D58	D58 Prototype- Sb_2S_3	Bi_2S_3 , Sb_2S_3
Zn_3P_2	D59	D59 Prototype- Zn_3P_2	Cd_3As_2 , Zn_3As_2 , Zn_3P_2
Ordered ternary s.s.	oD59	D59 Prototype- Zn_3P_2	Ordered ternary s.s.
Cr_3C_2	D510	D510 Prototype- Cr_3C_2	Cr_3C_2
Al_3Ni_2	D513	D513 Prototype- Al_3Ni_2	Al_3Ni_2
Si_2Zr_3	D5a	D5a Prototype- Si_2U_3	Si_2Zr_3
Al_4C_3	D71	D71 Prototype- Al_4C_3	Al_4C_3
$(\text{Ag}, \text{Cu})_5\text{Zn}_8$, $\gamma\text{-AlCu}$	D82	D82 Prototype- Cu_5Zn_8	$(\text{Ag}, \text{Cu}, \text{Fe})_5\text{Zn}_8$, $\gamma\text{-AlCu}$, $\gamma\text{-NiZn}$, $(\text{Ag}, \text{Cu})_5\text{Cd}_8$, $\varepsilon\text{-Al}_8(\text{Cr}, \text{Fe}, \text{V})_5$
$\text{V}_6\text{Ga}_7(\text{ht})$	D82a	D82 Prototype- Cu_5Zn_8	$\text{V}_6\text{Ga}_7(\text{ht})$
Cu_9Al_4 , InAg_2	D83	D83 Prototype- Cu_9Al_4	$\gamma\text{-Cu}_9\text{Al}_4$, $\gamma\text{-Co}_9\text{Zn}_4$, In_5Ag_8
Cr_{23}C_6	D84	D84 Prototype- Cr_{23}C_6	Cr_{23}C_6 , Mn_{23}C_6
$\mu\text{-Fe}_7\text{W}_6$	D85	D85 Prototype- Fe_7W_6 (Frank-Kasper)	$(\text{Co}, \text{Fe})_7(\text{Mo}, \text{Nb}, \text{W})_6$
$\varepsilon\text{-Cu}_{15}\text{Si}_4$	D86	D86 Prototype- $\text{Cu}_{15}\text{Si}_4$	$\varepsilon\text{-Cu}_{15}\text{Si}_4$
Mn_5Si_3 , Ti_5Si_3 , Ti_5Sn_3	D88	D88 Prototype- Mn_5Si_3	Cr_5Si_3 , Fe_5Si_3 , Mn_5Si_3 , Nb_5Si_3 , Ti_5Si_3 , V_5Si_3 , W_5Si_3 , Zr_5Si_3 , Ti_5Sn_3 , Zr_5Sn_3
(Fe-Ni) pentlandite, Co_9S_8	D89	D89 Prototype- Co_9S_8	(Fe-Ni) pentlandite, Co_9S_8
Al_8Mn_5 , Al_8Cr_5	D810	D810 Prototype- Al_8Cr_5	Al_8Mn_5 , Al_8Cr_5
$\text{Co}_{23}\text{Zr}_6$	D8a	D8a Prototype- $\text{Mn}_{23}\text{Th}_6$	$\text{Co}_{23}\text{Zr}_6$
$\sigma\text{-FeCr}$, $\sigma\text{-AlNb}$, $\sigma\text{-MnV}$, $\sigma\text{-NiV}$, $\sigma\text{-FeMo}$	D8b	D8b Prototype- FeCr	$\sigma\text{-FeCr}$, intermetallic phases in Al-Mo-Nb and Mn-Ni-V-W
Al_9Co_2	D8d	D8d Prototype- Al_9Co_2	Al_9Co_2
T-phase, Tau $\text{Mg}_{32}(\text{Al}, \text{Zn})_{49}$	D8e	Tau Prototype- $\text{Mg}_{32}(\text{Al}, \text{Zn})_{49}$	Ternary phases in Mg-Al-Cu-Zn-(Ag)
$\text{Mg}_5(\text{Ga}, \text{In})_2$	D8g	D8g Prototype- Ga_2Mg_5	$\text{Mg}_5(\text{Ga}, \text{In})_2$

Ca_5X_3 D8I	D8I	D8I Prototype-Cr5B3	Ca_5Si_3 , Ca_5Sn_3 , Ca_5Zn_3 , Ca_5Ag_3
Cr_5Si_3	D8m	D8m Prototype-W5Si3	$(\text{Cr}, \text{Mo}, \text{Nb}, \text{V}, \text{W})_5\text{Si}_3$
CuFeS2	E11	E11 Prototype-CuFeS2	chalcopyrite
FeZr_3 , CoZr_3	E1a	E1a Prototype-FeZr3	$(\text{Co}, \text{Fe})\text{Zr}_3$
$(\text{Fe}, \text{Mn}, \text{Ti})_3\text{AlC}$	E21	E21 Prototype-CaTiO3	$(\text{Fe}, \text{Mn}, \text{Ti})_3\text{AlC}$
$\text{Al}_7\text{Cu}_2\text{Fe}$, Al_7CuMn_2	E9a	E9a Prototype-Al7Cu2Fe	$\text{Al}_7\text{Cu}_2\text{Fe}$, Al_7CuMn_2 , $\tau_9\text{-Al}_7\text{Cu}_2\text{Zr}$
$(\text{Co}, \text{Fe}, \text{Ni})_3\text{W}_3\text{C}$	E93	E93 Prototype-Fe3W3C	$(\text{Co}, \text{Fe}, \text{Ni})_3\text{W}_3\text{C}$
$(\text{Co}, \text{Ni})_3\text{S}_4$, NiFe_2S_4	H11	H11 Prototype-Fe3O4	Ni_3S_4 (polydymite), NiFe_2S_4 (violarite), Co_3S_4 (linnaeite)
$\gamma\text{-AlTi}$, MgIn_n , L1 ₀	L10	L10 Prototype-AuCu	AlTi, MgIn, GaTi
CuTi L1 ₁	L11	L11 Prototype-CuTi	CuTi
Al_3M L1 ₂	L12	L12 Prototype-AuCu3	Pt ₃ Ti
$\text{Ni}_3(\text{Al}, \text{Fe}, \text{Si})$ L1 ₂	L12c	L12-FCC!FCC-A1	Ag_3Mg , Ni_3Al , Ni_3Fe , Ni_3Si , Si_3Ni
MgIn_3 , InMg_3 L1 ₂	L12d	L12d Prototype-AuCu3	MgIn ₃ , InMg ₃ , InAg ₃
Pt ₃ Ga, Pt ₃ Ge, Ni ₃ Ga	L12e	L12e Prototype-AuCu3	Pt ₃ Ga, Pt ₃ Ge, Ni ₃ Ga
AlCu ₂ Mn, AlNi ₂ Ti	L21	Heusler Prototype-AlCu ₂ Mn	AlCu ₂ Mn, AlNi ₂ Ti
TiH ₂ (low-T), ZrH ₂	L'2	L'2 Prototype-TiH2	TiH ₂ (low-T), ZrH ₂ , SiPt ₂
AgMg ₃	AMg3	cF264 Prototype-AgMg3	AgMg ₃
Al ₁₁ Mn ₄ (low-T)	aP15	aP15 Prototype-Al11Mn4	Al ₁₁ Mn ₄ (low-T), Al ₁₁ Cr ₄
$\xi\text{-Al}_2\text{Fe}$	aP18	aP18 Prototype-Al2Fe	$\xi\text{-Al}_2\text{Fe}$
$(\text{Al}, \text{In})_3\text{Ca}_8$	aP22	aP22 Prototype-Ca8In3	$(\text{Al}, \text{In})_3\text{Ca}_8$
Cu ₇ In ₃	aP40	aP40 Prototype-Cu7In3	Cu ₇ In ₃
Cu₂S, Cu₂Se	cF44	cF44 Prototype-Cu2Se	Cu ₂ S(digenite/bornite), Cu ₂ Se(berzelianite), gamma-Ag ₂ S, beta1-Ni ₂ Te(ht)
Ti ₂ Co, Ti ₂ Ni	cF96	cF96 Prototype-Ti2Ni	Ti ₂ Co, Ti ₂ Ni
Argentite-Ag₂S	cI20	cI20 Prototype-Ag2S	Ag ₂ S, Ag ₂ Se
Al ₁₂ Mn, Al ₁₂ W	cI26	cI26 Prototype-Al12W	Al ₁₂ (Mn, Mo, W) (Al ₁₂ Fe-metastable)
Ni ₂ S-β ₂	cP_	beta2	Ni ₂ S-beta2
CaAlSi, SrAlSi	hP3_	hP3 Prototype-AlB2	CaAlSi, SrAlSi
δ-CuZn	hP3	CuZn_delta	δ-CuZn, $\zeta\text{-AgCd}$, $\varepsilon\text{-AgCd}_3$
Fe ₂ Si	hP6	hP6 Prototype-Ni2Al	Ni ₂ Al, Fe ₂ Si
M ₂ AlC	hP8c	MAX Prototype-Cr2AlC	M ₂ AlC ($M = \text{Cr}, \text{Ti}, \text{Nb}, \text{V}$)
Bi ₂ Te ₃	hP9	hP9 Prototype-Sb2Te	Bi ₂ Te ₃
δ-WC, η-MoC	hP12	eta-MoC	δ-WC, η-MoC
Cu ₂ S chalcocite(ht)	hP16	hP16 Prototype-Cu2S	Cu ₂ S chalcocite(ht)
Ti ₅ Ga ₄	hP18	hP18 Prototype-Ti5Ga4	Ti ₅ Ga ₄
β-Ti ₆ Sn ₅ , V ₆ Ga ₅	h22P	hP22 Prototype Ti6Sn5	β-Ti ₆ Sn ₅ , V ₆ Ga ₅

β -AlMnSi	hP26	AlMnSi_beta	β -AlMnSi
SbSn	hR3	hR3 Prototype-SbSn"	SbSn
Bi ₂ Te ₃	hR15	hR15 Prototype-Bi ₂ Te ₃	Bi ₂ Te ₃
(Al,Ga,In) ₂ Li ₃	h15R	hR15 Prototype-Al ₂ Li ₃	(Al,Ga,In) ₂ Li ₃
V ₄ C ₃	hR21	hR21 Prototype-Sn ₄ P ₃	V ₄ C ₃
Cr ₂ S ₃	hR30	hR30 Prototype-Cr ₂ S ₃	Cr ₂ S ₃
Al ₅ Mo	h36R	h36R Prototype-MoAl ₅	Al ₅ Mo
Mg ₃ In	hR48	hR48 Prototype-Mg ₃ In	Mg ₃ In
Ni ₇ Zr ₂ , Co ₇ Nb ₂	mC18	mC18 Prototype-Ni ₇ Zr ₂	Ni ₇ Zr ₂ , Co ₇ Nb ₂
Mo ₃ Al ₈	mC22	mC22 Prototype-Mo ₃ Al ₈	Mo ₃ Al ₈
ζ -Zn ₁₃ Fe, γ_2 -Zn ₁₃ Co	mC28	mC28 Prototype-Zn ₁₃ Co	ζ -Zn ₁₃ Fe, γ_2 -Zn ₁₃ Co
Ni ₅ Sb ₂	m28C	mS28 Prototype-Ni ₅ Sb ₂	Ni ₅ Sb ₂
Al ₄ Mo, Al ₄ W	mC30	mC30 Prototype-Al ₄ W	Al ₄ Mo, Al ₄ W
β -AlFeSi (τ_6)	mC52	AlFeSi_beta	β -AlFeSi (τ_6)
(Ag,Cu)P ₂	mp12	mp12 Prototype-CuP ₂	(Ag,Cu)P ₂
Acanthite-Ag ₂ S	m12P	m12P Prototype-Ag ₂ S	Acanthite-Ag ₂ S
ZnAs ₂	mp24	mp24 Prototype-ZnAs ₂	ZnAs ₂ , β -ZnP ₂
'Al ₃ Fe'	m102	mC102 Prototype-Al ₁₃ Fe ₄	Al ₁₃ Fe ₄ , Al ₁₃ Co ₄
Mg ₂ Zn ₃	m110	Mg ₂ Zn ₃	Mg ₂ Zn ₃
CrS	mS8	mS8 Prototype-CrS	CrS
Ni ₃ Sn ₄	mC14	mS14 Prototype-Ni ₃ Sn ₄	Ni ₃ Sn ₄
Cr ₃ S ₄	mS14	mS14 Prototype-Cr ₃ S ₄	Cr ₃ S ₄
Pt ₃ Ge, Pt ₃ Si _(lt)	mS16	mS16 Prototype-Pt ₃ Ge	Pt ₃ (Ge,Si)
η_2 -AlCu, θ -InCu	mS20	eta2 Prototype-AlCu	η_2 -AlCu, θ -InCu
GeAs, SiAs	mS24	mS24 Prototype-SiAs	(Ge,Si)(As,P)
Hägg χ -Fe ₅ C ₂	mS28	Haegg Prototype-Mn ₅ C ₂	χ -Fe ₅ C ₂ , Mn ₅ C ₂
MoAl ₃	mS32	mS32 Prototype-MoAl ₃	MoAl ₃
ζ' -Ni ₁₃ In ₉	mS44	mS44 Prototype-Ni ₁₃ Ga ₉	ζ' -Ni ₁₃ In ₉
Al ₁₄ Ca ₁₃	mS54	mS54 Prototype-Al ₁₄ Ca ₁₃	Al ₁₄ Ca ₁₃
Al ₃ Hf ₂ , Al ₃ Zr ₂ , Ga ₃ Zr ₂	oF40	oF40 Prototype-Al ₃ Zr ₂	Al ₃ Hf ₂ , Al ₃ Zr ₂ , Ga ₃ Zr ₂
(Cr,Mo,V)Ni ₂	oI6	oI6 Prototype-MoPt ₂	(Cr,Mo,V)Ni ₂
(Nb,Ti) ₆ Sn ₅	oI44	oI44 Prototype-Ti ₆ Sn ₅	(Nb,Ti) ₆ Sn ₅
α -(Pd,Pt)Ti	oP4	oP4 Prototype-AuCd	α -(Pd,Pt)Ti
Ag ₃ Sb	oP4a	oP4a Prototype-Ag ₃ Sb	Ag ₃ Sb
CaMgSi	oP12	oP12 Prototype-TiNiSi	CaMgSi
naumannite-Ag ₂ Se	o12P	oP12 Prototype-Ag ₂ Se	naumannite-Ag ₂ Se
Cu ₄ Ti	oP20	oP20 Prototype-ZrAu ₄	β' -Cu ₄ Ti
(Co,Mn,Ni) ₃ Sn ₂	o20P	oP20 Prototype-Ni ₃ Sn ₂	(Co,Mn,Ni) ₃ Sn ₂
SiAs ₂ , GeAs ₂	oP24	oP24 Prototype-GeAs ₂	(Ge,Si)As ₂

Ca_7Ge_6 , Ca_7Sn_6	oP52	oP52 Prototype- Ca_7Sn_6	$\text{Ca}_7(\text{Ge},\text{Sn})_6$
δ -NiMo	oP56	NiMo-delta	δ -NiMo
GaCa, GeCa, NiZr	oS8	oS8 Prototype-TII	GaCa, GeCa, NiZr
ZrGa_2	oS12	oS12 Prototype- ZrGa_2	ZrGa_2
AgCuS stromayerite	o12S	oS12	AgCuS
$(\text{Pt},\text{Ni})_5\text{Ga}_3$	o16S	oS16 Prototype- Pt_5Ga_3	$(\text{Pt},\text{Ni})_5\text{Ga}_3$
$\text{Zr}_3(\text{Ga},\text{Ge},\text{Si})_5$	oS32	oS32 Prototype- Pu_3Pd_5	$\text{Zr}_3(\text{Ga},\text{Ge},\text{Si})_5$
Li_7Sn_2	oS36	oS36 Prototype- Li_7Ge_2	Li_7Sn_2
β -FeSi ₂	oS48	Luobusaite Prototype- FeSi_2	β -FeSi ₂
α -Ni ₇ S ₆	oS56	oS56 Prototype-Ni ₇ S ₆	Ni ₇ S ₆
'Ni ₁₀ Zr ₇ '	oS68	oS68 Prototype- $\text{Zr}_7\text{Ni}_{10}$	'Ni ₁₀ Zr ₇ '
Godlevskite Ni ₉ S ₈	oC68	oC68 Prototype-Ni ₉ S ₈	Ni ₉ S ₈
(Cd,Zn)As ₂	tI2I	tI12 Prototype-CdAs ₂	(Cd,Zn)As ₂
η -Al ₂ Ti	tI24	tI24 Prototype-HfGa ₂	η -Al ₂ Ti, Ga ₂ Ti, Al ₂ Mg-metastable
ε -Al ₃ Ti(lt)	tI32	tI32 Prototype-TiAl ₃	ε -Al ₃ Ti(lt)
CaZn ₁₁	tI48	tI48 Prototype-BaCd11	CaZn ₁₁
(Cd,Zn) ₃ (As) ₂	i160	tI160 Prototype- Cd ₃ As ₂	(Cd,Zn) ₃ (As) ₂
α -FeSi _{2(ht)}	tP3	Ferdasilicate Prototype-FeSi ₂	α -FeSi ₂ (high-T)
Pt ₃ Ga _(ht) , Pt ₃ Al	tP16	tP16 Prototype-Pt ₃ Ga	Pt ₃ Ga _(ht) , Pt ₃ Al
Al ₂ Zr ₃	tP20	tP20 Prototype-Al ₂ Zr ₃	Al ₂ Zr ₃ , Zn ₂ Zr ₃
(Cd,Zn)P ₂	tP24	tP24 Prototype-ZnP ₂	(Cd,Zn)P ₂
ζ -Al ₅ Ti ₂	tP28	tP28 Prototype Al ₅ Ti ₂	ζ -Al ₅ Ti ₂
Ti ₃ P, Ti ₃ Si	tP32	tP32 Prototype-Ti ₃ P	M_3 P (with $M = \text{Ti}, \text{V}, \text{Zr}$) $X_3\text{Si}$ (with $X = \text{Nb}, \text{Ti}, \text{Zr}$)
(Ti,Zr) ₅ Si ₄	tP36	tP36 Prototype-Zr ₅ Si ₄	(Ti,Zr) ₅ Si ₄
(Cd,Zn) ₃ (As) ₂	p160	tP160 Prototype- Zn ₃ As ₂	(Cd,Zn) ₃ (As) ₂
β -Al ₃ Mg ₂	Beta	Beta Prototype-Mg ₂₈ Al ₄₅	β -Al ₃ Mg ₂
η -Al ₅ Fe ₂	Eta	Eta Prototype-Al ₅ Fe ₂	η -Al ₅ Fe ₂
ϕ -Mg ₆ (Al,Zn) ₅	Phi	Phi AIMgZn	ϕ -Mg ₆ (Al,Zn,Cu) ₅
ζ -AlCu or ζ -Al ₉ Cu ₁₁	Zeta	Zeta Prototype-AlCu	ζ -Al ₉ Cu ₁₁
E -phase Al ₁₈ Cr ₂ Mg ₃ Zn ₂₂ Zr, CaCr ₂ Al ₂₀	E	E Prototype-CeCr ₂ Al ₂₀	Al ₁₈ Cr ₂ Mg ₃ , Al ₁₈ Mn ₂ Mg ₃ , Al ₁₈ Mo ₂ Mg ₃ , Al ₁₈ Ti ₂ Mg ₃ , Al ₁₈ V ₂ Mg ₃ , Al ₁₈ W ₂ Mg ₃ CaCr ₂ Al ₂₀ , Zn ₂₂ Zr
Q-Al ₃ Cu ₂ Mg ₉ Si ₇	Q	Q Prototype-Th ₇ S ₁₂	Al ₃ Cu ₂ Mg ₉ Si ₇
P phase	P	P phase	CrMoNi ternary solution
P1	P1	M3Ca2	(Mg,Cu) ₃ Ca ₂

P2	P2	MCu3	(Mg,Ca)Cu ₃
R phase	R	R phase	(Co,Fe,Ni) ₂₇ W ₁₄ (Co,Cr,Fe,Ni,W) ₁₂
S-Al ₂ CuMg	S	S Prototype-Al ₂ CuMg	S-Al ₂ CuMg (ordered E1a)
μ -Al ₄ Mn, Al ₄ Cr	Al4M	Al4Mn	Al ₄ Mn, Al ₄ Cr (metastable Al ₄ Fe)
'Al ₃ Mn(ht)'	A3M_	oP156 Prototype-Al11Mn4	Al ₂₉ Mn ₁₀ (high-T)
Al ₁₁ Cr ₂	AlCr	Al11Cr2	Al ₁₁ Cr ₂
α -AlFeSi, τ_5	Tau5	AlFeSi_alpha	stoichiometry ~Al ₇ Fe ₂ Si (chinese script/rods)
γ -AlFeSi, τ_2	Tau2	AlFeSi_gamma	stoichiometry τ_2 ~Al ₃ FeSi
α -AlMnSi	AlM1	AlMnSi_alpha	stoichiometry τ_9 ~Al ₉ Mn ₂ Si
α -AlFeMnSi	AFMS	AlFeMnSi_alpha	stoichiometry ~Al ₁₈ (Fe,Mn) ₄ Si ₃
'MgZn' or Mg ₁₂ Zn ₁₃	MgZn	Mg12Zn13	Mg ₁₂ Zn ₁₃
δ -AlCu or δ -Al ₂ Cu ₃	Dlta	AlCu_delta	δ -Al ₂ Cu ₃
η_1 -AlCu	EtaH	Eta1 Prototype-AlCu	η_1 -AlCu
γ -AlCu(high-T)	Gam5	gamma Prototype-AlCu	γ -Al ₄ Cu ₉ (high-T)
τ -AlCuZn	Tau_	AlCuZn_Tau	τ -Al ₉ Cu ₉ Zn ₂ (approx.)
β -Zn ₃ Li ₂	Li2Z	Zn3Li2	β -Li ₂ Zn ₃
σ -CrMn	HSig	High_Sigma	~Cr ₂ Mn ₇
Ca(Mg,Zn) ₅	IM1	IM1-CaMgZn	CaMg ₄ Zn, CaMgZn ₄
(Li,Mg) ₁₇ Sn ₄	L17X	cF420 Prototype-Li17Pb4	Li ₁₇ Sn ₄
γ -AgMg ₄	hP_G	AgMg4 gamma_hp	γ -AgMg ₄
Ag ₁₇ Mg ₅₄	A17M	oP142 Prototype-Hf54Os17	Ag ₁₇ Mg ₅₄
Li ₂ Ag	L2A	Prototype-Li2Ag	~Li ₂ Ag
Li ₃ Ag	L3A	Prototype-Li3Ag	~Li ₃ Ag
Li ₆ Ag	L6A	Prototype-Li6Ag	~Li ₆ Ag
(Co,Fe) ₆ W ₆ C	M12C	cF104 Prototype-Fe6W6C	(Co,Fe) ₆ W ₆ C
Mn ₁₁ Si ₁₉	NCL1	tP20 Prototype-Mn11Si19	Mn ₁₁ (Si,Al) ₁₉ Nowotny Chimney Ladder Phase
Al ₆₃ Mo ₃₇	AM8	AM8 Al63Mo37	Al ₆₃ Mo ₃₇
"As ₃ P ₂ " or "AsP"	AsP	AsP-orthorhombic	~"As ₃ P ₂ "
γ_1 -FeZn	Gam1	FeZn-Gamma-1	Fe ₇ (Fe,Zn) ₆ Zn ₃₈
δ_1 -FeZn	FeZ2	FeZn-delta-1	Fe(Al,Fe,Zn) ₃ Zn ₁₃
(Cu,Ni)S₂	Vill	Villamaninite	(Cu,Ni)S ₂
Heazlewoodite-Ni₃S₂	Heaz	Heaz Prototype-Ni3S2	Ni ₃ S ₂
Chalcocite-Cu₂S(lt)	Chal	Chalcocite	Cu ₂ S(lt)
Cu_{1.95}S(djurleite)	Djur	Djurleite	Cu _{1.95} S
Iss-CuMeS_{2-x}	CuMS	iss_CuMeS2-x	non-stoichiometric solid solution [Cu,Ni,Fe] ₂ S
(Fe,Ni) ₃ Te ₂	M3T2	M3T2	(Fe,Ni) ₃ Te ₂
M ₄ AlC ₃	MAX2	MAX2 Prototype-CsYb3Se4	(Cr,Nb,Ti,V)AlC ₃

Table 3: Pure compounds and their phases in the FScopp 8.3 Database

FScopp83 database 1077 compounds, 1442 phases							
		S1	S2	S3	S4	S5	L
Ag	FScopp83	S1	S2	S3	S4	S5	L
Ag11Hg15	FScopp83	S					
Ag11Hg9	FScopp83	S					
Ag2Ca	FScopp83	S					
Ag2Ga	FScopp83	S					
Ag2O	FScopp83	S					
Ag2S	FScopp83	S1	S2	S3			
Ag2Se	FScopp83	S1	S2				
Ag2Te	FScopp83	S1	S2	S3			
'Ag3Be8'	FScopp83	S					(Ag) 2.97 (Be) 8.03
Ag3Ca5	FScopp83	S					
Ag3CuS2	FScopp83	S					
Ag3Ga2	FScopp83	S					
Ag3P11	FScopp83	S					
Ag3Sb	FScopp83	S					
Ag5Cu3S4	FScopp83	S					
Ag5Te3	FScopp83	S1	S2				
Ag7Ca2	FScopp83	S					
Ag8SnS6	FScopp83	S1	S2				
Ag9Ca2	FScopp83	S					
AgCa	FScopp83	S					
AgCa3	FScopp83	S					
AgCd	FScopp83	S					
AgCu4Zr	FScopp83	S					
AgCuS	FScopp83	S					
AgMg4	FScopp83	S					
AgP2	FScopp83	S					
AgTi2	FScopp83	S					
AgZr	FScopp83	S					
AgZr2	FScopp83	S					
Al	FScopp83	S1	S2	S3	S4	S5	S6
Al10V	FScopp83	S					
Al116Cr63Si21	FScopp83	S					
Al11Cr2	FScopp83	S					
Al11Cu5Mn3	FScopp83	S					
Al11Fe3Si6	FScopp83	S					
Al11Mn3Zn2	FScopp83	S					
Al11Mn4	FScopp83	S					
Al12Mn	FScopp83	S					
Al12Mo	FScopp83	S					
Al12W	FScopp83	S					
Al13Co4	FScopp83	S					
Al13Cr4Si4	FScopp83	S					
Al14Ca13	FScopp83	S					
Al14Fe3Si3	FScopp83	S					
Al15Cu8Li2	FScopp83	S					
Al17Mo4	FScopp83	S					
Al18Cr2Mg3	FScopp83	S					
Al18Mg3Mn2	FScopp83	S					
Al18Mg3Mo2	FScopp83	S					
Al18Mg3Ti2	FScopp83	S					
Al18Mg3V2	FScopp83	S					
Al18Mg3W2	FScopp83	S					
Al21Pt5	FScopp83	S					
Al21Pt8	FScopp83	S					
Al22Mo5	FScopp83	S					
Al23CuFe4	FScopp83	S					
Al23V4	FScopp83	S					
Al2Au	FScopp83	S					
Al2Ca	FScopp83	S1	S2				

A12CaZn2	FScopp83	S				
A12Cu	FScopp83	S1	S2	S3		
A12Cu3	FScopp83	S				
A12CuLi	FScopp83	S				
A12CuMg	FScopp83	S				
A12Fe	FScopp83	S				
A12Li	FScopp83	S1	S2			
A12Li18Si6	FScopp83	S				
A12Li3	FScopp83	S				
A12Mn2Si3	FScopp83	S				
A12O3	FScopp83	S1	S2	S3	S4	
A12Pt	FScopp83	S				
A12S3	FScopp83	S				
A12Se3	FScopp83	S				
A12Ti	FScopp83	S1	S2			
A12W	FScopp83	S				
A12Zr	FScopp83	S				
A12Zr3	FScopp83	S				
A130Mg23	FScopp83	S				
A139Cu33Zr6	FScopp83	S				
A13Ca8	FScopp83	S				
A13Co	FScopp83	S				
A13Cr	FScopp83	S1	S2			
A13Cu	FScopp83	S				
A13Cu2	FScopp83	S				
A13Li	FScopp83	S1	S2	S3		
A13Li15Si6	FScopp83	S				
A13Li7Si4	FScopp83	S				
A13Mg	FScopp83	S				
A13MnSi2	FScopp83	S				
A13Mo	FScopp83	S				
A13Nb	FScopp83	S				
A13Ni	FScopp83	S				
A13Ni5	FScopp83	S				
A13Pt2	FScopp83	S				
A13Pt5	FScopp83	S				
A13Ti	FScopp83	S1	S2	S3	S4	
A13V	FScopp83	S1	S2	S3		
A13Zr	FScopp83	S1	S2	S3		
A13Zr2	FScopp83	S				
A13Zr5	FScopp83	S				
A14C3	FScopp83	S				
A14C4Si	FScopp83	S				
A14Ca	FScopp83	S				
A14Cr	FScopp83	S				
A14Fe	FScopp83	S				
A14Li9	FScopp83	S				
A14Mn	FScopp83	S				
A14Mo	FScopp83	S				
A14W	FScopp83	S				
A14Zr5	FScopp83	S				
A153Mg14Li33	FScopp83	S				
A157Cu11Li32	FScopp83	S				
A15Co2	FScopp83	S				
A15Cu2Mg8Si6	FScopp83	S				
A15Cu6Mg2	FScopp83	S				
'A15CuLi3'	FScopp83	S				(Al) 4.95 (Cu) 1.053 (Li) 2.997
A15Li	FScopp83	S				
A15Mo	FScopp83	S				
A15W	FScopp83	S				
A163Mo37	FScopp83	S				
A16Fe	FScopp83	S				
A16Mn	FScopp83	S				
A16Ni3Si	FScopp83	S				
A177W23	FScopp83	S				

Al7Cr	FScopp83	S
Al7Cu2Fe	FScopp83	S
Al7Cu2Zr	FScopp83	S
Al7Cu3Mg6	FScopp83	S
Al7CuFe2	FScopp83	S
Al7CuMn2	FScopp83	S
Al7CuZr2	FScopp83	S
Al7W3	FScopp83	S
Al8C7Si	FScopp83	S
Al8Cr5	FScopp83	S1 S2
Al8CrTi3	FScopp83	S
Al8FeMg3Si6	FScopp83	S
Al8Mo3	FScopp83	S
Al8V5	FScopp83	S
Al99Mn23	FScopp83	S
Al9Co2	FScopp83	S
Al9Cr3Si	FScopp83	S
Al9Cr4	FScopp83	S1 S2
Al9Cu11	FScopp83	S
AlAs	FScopp83	S
AlAu	FScopp83	S
AlAu2	FScopp83	S
AlAu4	FScopp83	S
AlCa5Zn2	FScopp83	S
AlCo	FScopp83	S
AlCr2	FScopp83	S
AlCu	FScopp83	S1 S2
AlCu2	FScopp83	S
AlCu3Mn2	FScopp83	S
AlH3	FScopp83	S
AlLi	FScopp83	S1 S2
AlLiSi	FScopp83	S
AlMg	FScopp83	S
AlMgAg	FScopp83	S
AlMo3	FScopp83	S
AlNaSi	FScopp83	S
AlP	FScopp83	S
AlPt	FScopp83	S
AlPt2	FScopp83	S1 S2
AlPt3	FScopp83	S
AlSb	FScopp83	S
AlSn2Zr5	FScopp83	S
AlTi	FScopp83	S
AlZr	FScopp83	S1 S2
AlZr2	FScopp83	S
AlZr3	FScopp83	S1 S2
As	FScopp83	S1 S2 S3 S4 S5 L
'As2Cu5'	FScopp83	S
As2S3	FScopp83	S
As2S5	FScopp83	S
As2Te3	FScopp83	S
As4S3	FScopp83	S1 S2
As4S4	FScopp83	S1 S2
AsCu8	FScopp83	S
AsIn	FScopp83	S
Au	FScopp83	S1 S2 S3 L
Au2Bi	FScopp83	S
Au2Pb	FScopp83	S
Au2Ti	FScopp83	S
Au3In	FScopp83	S
Au3Sn12Co5	FScopp83	S
'Au4Zn5'	FScopp83	S
Au5Zn3	FScopp83	S
Au7Ga2	FScopp83	S
Au7Ga3	FScopp83	S

(As) 2.058 (Cu) 4.942

(Au) 3.96 (Zn) 5.04

Au7In3	FScopp83	S				
Au8Al3	FScopp83	S				
'Au8Ga2'	FScopp83	S				(Au) 7.89476 (Ga) 2.10526
AuGa	FScopp83	S				
AuGa2	FScopp83	S				
AuIn	FScopp83	S				
AuIn2	FScopp83	S				
AuSb2	FScopp83	S				
AuSn	FScopp83	S				
AuSn2	FScopp83	S				
AuSn4	FScopp83	S				
AuTe2	FScopp83	S				
AuTi	FScopp83	S1	S2			
AuTi3	FScopp83	S				
AuZn3	FScopp83	S				
Be	FScopp83	S1	S2	S3	L	
Bi	FScopp83	S1	S2	S3	S4	L
Bi2S3	FScopp83	S				
Bi2Te	FScopp83	S				
Bi3In5	FScopp83	S				
Bi3Ni	FScopp83	S				
Bi4Te3	FScopp83	S				
BiIn	FScopp83	S				
BiIn2	FScopp83	S				
BiLi	FScopp83	S				
BiNa	FScopp83	S				
C	FScopp83	S1	S2	L		
Ca	FScopp83	S1	S2	S3	L	
Ca11Ga7	FScopp83	S				
Ca14Si19	FScopp83	S				
Ca25Ga59	FScopp83	S				
Ca28Ga11	FScopp83	S				
Ca2Cu	FScopp83	S				
Ca2Ge	FScopp83	S				
Ca2Mg55Zn43	FScopp83	S				
Ca2Mg5Zn13	FScopp83	S				
Ca2Ni7	FScopp83	S				
Ca2Pb	FScopp83	S				
Ca2Si	FScopp83	S				
Ca2Sn	FScopp83	S				
Ca3Ga5	FScopp83	S				
Ca3Ga8	FScopp83	S				
Ca3Mg3Zn14	FScopp83	S				
Ca3P2	FScopp83	S				
Ca3Si4	FScopp83	S				
Ca3Zn	FScopp83	S				
Ca4Al3Mg	FScopp83	S				
Ca5Ga3	FScopp83	S				
Ca5Ge3	FScopp83	S				
Ca5Pb3	FScopp83	S				
Ca5Si3	FScopp83	S				
Ca5Zn3	FScopp83	S				
Ca7Ge6	FScopp83	S				
Ca7Mg6Si14	FScopp83	S				
Ca7Sn6	FScopp83	S				
CaAl2Si2	FScopp83	S				
CaAlH5	FScopp83	S				
CaBe13	FScopp83	S				
CaC2	FScopp83	S1	S2	S3	S4	L
CaCr2Al120	FScopp83	S				
CaCu	FScopp83	S				
CaCu5	FScopp83	S				
CaGa	FScopp83	S				
CaGa2	FScopp83	S1	S2			
CaGa4	FScopp83	S1	S2			

CaGe2	FScopp83	S
CaH2	FScopp83	S
CaLi2	FScopp83	S
CaMgSi	FScopp83	S
CaNi2	FScopp83	S
CaNi3	FScopp83	S
CaNi5	FScopp83	S
CaO	FScopp83	S
CaPb	FScopp83	S
CaPb3	FScopp83	S
CaS	FScopp83	S
CaSi	FScopp83	S
CaSi2	FScopp83	S
CaTi2Al20	FScopp83	S
CaZn	FScopp83	S
CaZn11	FScopp83	S
CaZn13	FScopp83	S
CaZn2	FScopp83	S
CaZn3	FScopp83	S
CaZn5	FScopp83	S
Cd	FScopp83	S1 S2 S3 S4 L
Cd10Cu3	FScopp83	S
Cd11Na2	FScopp83	S
Cd2Na	FScopp83	S
Cd3As2	FScopp83	S1 S2 S3 S4
Cd3Cu4	FScopp83	S
Cd3In	FScopp83	S
Cd3P2	FScopp83	S1 S2 S3
Cd5Ni	FScopp83	S
Cd6P7	FScopp83	S
Cd7P10	FScopp83	S
Cd8Cu5	FScopp83	S
CdAs2	FScopp83	S
CdCu2	FScopp83	S
CdNi	FScopp83	S
CdP2	FScopp83	S1 S2
CdP4	FScopp83	S
CdS	FScopp83	S
CdSb	FScopp83	S
CdTi	FScopp83	S
CdTi2	FScopp83	S
Co	FScopp83	S1 S2 S3 S4 S5 L
Co11Zr2	FScopp83	S
Co16Nb9	FScopp83	S
Co23C6	FScopp83	S
Co23Zr6	FScopp83	S
Co2Nb	FScopp83	S
Co2P	FScopp83	S
Co2Si	FScopp83	S
Co2Ti	FScopp83	S1 S2 S3
Co2Zr	FScopp83	S
Co3C2	FScopp83	S
Co3Mo	FScopp83	S
Co3Nb	FScopp83	S
Co3S4	FScopp83	S
Co3Si	FScopp83	S
Co3Sn2	FScopp83	S1 S2
Co3V	FScopp83	S
Co3W	FScopp83	S
Co4Zn9	FScopp83	S
Co6W6C	FScopp83	S
Co7C3	FScopp83	S
Co7Nb2	FScopp83	S
Co9S8	FScopp83	S
CoGa3	FScopp83	S

CoIn2	FScopp83	S							
CoIn3	FScopp83	S							
CoO	FScopp83	S							
CoP	FScopp83	S							
CoS	FScopp83	S							
CoS2	FScopp83	S							
CoSb	FScopp83	S							
CoSb2	FScopp83	S							
CoSb3	FScopp83	S							
CoSe2	FScopp83	S							
CoSi	FScopp83	S							
CoSi2	FScopp83	S							
CoSn	FScopp83	S							
CoSn2	FScopp83	S							
CoTe	FScopp83	S							
CoTe2	FScopp83	S							
CoTi2	FScopp83	S							
CoV3	FScopp83	S							
CoZn7	FScopp83	S							
CoZr2	FScopp83	S							
CoZr3	FScopp83	S							
Cr	FScopp83	S1	S2	S3	S4	S5	S6	L	
Cr23C6	FScopp83	S							
Cr2AlC	FScopp83	S							
Cr2C	FScopp83	S							
Cr2O3	FScopp83	S							
Cr2P	FScopp83	S							
Cr2S3	FScopp83	S							
Cr2Ti	FScopp83	S1	S2						
Cr2Zr	FScopp83	S1	S2	S3					
Cr3C	FScopp83	S							
Cr3C2	FScopp83	S							
Cr3Ga	FScopp83	S1	S2						
Cr3Mn5	FScopp83	S							
Cr3P	FScopp83	S							
Cr3S4	FScopp83	S1	S2						
Cr3Si	FScopp83	S							
'Cr4S6'	FScopp83	S							(Cr) 4.1 (S) 5.9
Cr5S6	FScopp83	S							
Cr5Si3	FScopp83	S1	S2	S3					
Cr7C3	FScopp83	S							
Cr7S8	FScopp83	S							
CrGa4	FScopp83	S							
Crs	FScopp83	S1	S2						
CrSi	FScopp83	S							
CrSi2	FScopp83	S							
CrZn13	FScopp83	S							
CrZn17	FScopp83	S							
Cu	FScopp83	S1	S2	S3	S4	L			
'Cu10Sn3'	FScopp83	S							(Cu) 9.997 (Sn) 3.003
Cu10Zr7	FScopp83	S							
Cu11Fe2S13	FScopp83	S							
Cu15Si4	FScopp83	S							
Cu16Mg6Si7	FScopp83	S							
Cu19Si6	FScopp83	S							
Cu2Mg	FScopp83	S1	S2	S3					
Cu2O	FScopp83	S							
Cu2P	FScopp83	S							
Cu2S	FScopp83	S1	S2	S3	S4				
Cu2Sb	FScopp83	S							
Cu2Se	FScopp83	S1	S2						
Cu2SiS3	FScopp83	S1	S2						
Cu2Te	FScopp83	S1	S2	S3	S4	S5	S6		
Cu2Ti	FScopp83	S							
Cu2TiZr	FScopp83	S							

Fe3O4	FScopp83	S1	S2	S3	S4			
Fe3P	FScopp83	S						
Fe3S4	FScopp83	S1	S2					
Fe3Si	FScopp83	S						
Fe3Sn2	FScopp83	S						
Fe5C2	FScopp83	S1	S2					
Fe5Si3	FScopp83	S1	S2					
Fe5Sn3	FScopp83	S1	S2	S3				
Fe60Te71	FScopp83	S						
Fe6W6C	FScopp83	S						
Fe7C3	FScopp83	S						
Fe7S6	FScopp83	S						
'Fe7S8'	FScopp83	S						(Fe) 7.016 (S) 8
Fe7W6	FScopp83	S						
Fe9S10	FScopp83	S						
Fe9S8	FScopp83	S1	S2					
FeBe12	FScopp83	S						
FeBe2	FScopp83	S						
FeBe5	FScopp83	S						
FeCr2S4	FScopp83	S						
FeP	FScopp83	S						
FeP2	FScopp83	S						
FeP4	FScopp83	S						
FeS	FScopp83	S1	S2					
FeS2	FScopp83	S1	S2					
FeSb	FScopp83	S						
FeSb2	FScopp83	S						
FeSe2	FScopp83	S						
FeSi	FScopp83	S1	S2	S3				
FeSi2	FScopp83	S1	S2	S3				
FeSn	FScopp83	S1	S2					
FeSn2	FScopp83	S						
FeSn2Zr6	FScopp83	S						
FeTe	FScopp83	S						
FeTe2	FScopp83	S1	S2					
FeTi	FScopp83	S						
Ga	FScopp83	S1	S2	S3	S4	S5	L	
Ga2Cu	FScopp83	S						
Ga2Pt	FScopp83	S1	S2					
Ga2Te3	FScopp83	S						
Ga2Te5	FScopp83	S						
Ga2Ti	FScopp83	S						
Ga2Zr	FScopp83	S						
Ga2Zr3	FScopp83	S						
Ga39Na22	FScopp83	S						
Ga3Pt2	FScopp83	S						
Ga3Pt5	FScopp83	S						
Ga3Te4	FScopp83	S						
Ga3Ti	FScopp83	S						
Ga3Ti2	FScopp83	S						
Ga3Ti5	FScopp83	S						
Ga3Zr	FScopp83	S						
Ga3Zr2	FScopp83	S						
Ga3Zr5	FScopp83	S						
Ga4Cr3	FScopp83	S						
Ga4Cu9	FScopp83	S						
Ga4Na	FScopp83	S						
Ga4Zr5	FScopp83	S						
Ga5Zr3	FScopp83	S						
Ga6Pt	FScopp83	S						
Ga6SnTe10	FScopp83	S						
Ga7Pt3	FScopp83	S						
Ga8Mo3	FScopp83	S						
GaAs	FScopp83	S						
GaCr	FScopp83	S						

GaP	FScopp83	S
GaPt	FScopp83	S
GaPt2	FScopp83	S
GaPt3	FScopp83	S1 S2 S3
GaSb	FScopp83	S
GaTe	FScopp83	S
GaZr	FScopp83	S1 S2
GaZr2	FScopp83	S
Ge	FScopp83	S1 S2 S3 L
GeAs	FScopp83	S
GeAs2	FScopp83	S
GeCa	FScopp83	S
GeP	FScopp83	S
GeTe	FScopp83	S
H	FScopp83	S L
Hg	FScopp83	S1 S2 S3 S4 L
Hg3Zn	FScopp83	S
Hg6Cu7	FScopp83	S
HgMg	FScopp83	S
HgMg3	FScopp83	S
HgPb2	FScopp83	S
HgS	FScopp83	S1 S2
HgSn38	FScopp83	S
HgSn4	FScopp83	S
HgSn7	FScopp83	S
HgZn2	FScopp83	S
HgZn3	FScopp83	S
In	FScopp83	S1 S2 S3 L
In2Ag	FScopp83	S
In2Ca	FScopp83	S
In2Li3	FScopp83	S
In2Pt	FScopp83	S
In2Pt3	FScopp83	S1 S2
In2Se3	FScopp83	S1 S2 S3 S4
In3Li13	FScopp83	S
In3Pt2	FScopp83	S
In4Se3	FScopp83	S
In5Na3	FScopp83	S
In5Pt6	FScopp83	S
In5Se7	FScopp83	S
In6Se7	FScopp83	S
In7Pt3	FScopp83	S
In9Na5	FScopp83	S
In9Pt13	FScopp83	S
In9Se11	FScopp83	S
InAg3	FScopp83	S
InCa3	FScopp83	S
InLi2	FScopp83	S
InLi6	FScopp83	S
InMg2	FScopp83	S
InNa	FScopp83	S
InNa2	FScopp83	S
InP	FScopp83	S
InPt	FScopp83	S
InPt2	FScopp83	S
InSb	FScopp83	S
InSe	FScopp83	S
Li	FScopp83	S1 S2 S3 L
Li12Si7	FScopp83	S
Li13Si4	FScopp83	S
Li22Si5	FScopp83	S
Li2Ga	FScopp83	S
Li2Ga7	FScopp83	S
Li2O	FScopp83	S
Li2Sb	FScopp83	S

Li3AlH6	FScopp83	S
Li3Bi	FScopp83	S
Li3Ga14	FScopp83	S
Li3Ga2	FScopp83	S
Li3Ga8	FScopp83	S
Li3Sb	FScopp83	S1 S2
Li5Ga4	FScopp83	S
Li5Ga9	FScopp83	S
Li7Si3	FScopp83	S
LiAlH4	FScopp83	S
LiGa	FScopp83	S
LiH	FScopp83	S
LiPb	FScopp83	S1 S2
LiZn	FScopp83	S
LiZn4	FScopp83	S
Mg	FScopp83	S1 S2 S3 S4 S5 S6 S7 L
Mg11Si7Al3	FScopp83	S
Mg12Zn13	FScopp83	S
Mg2Ca	FScopp83	S1 S2
Mg2Cu	FScopp83	S1 S2
Mg2Ga	FScopp83	S
Mg2Ga5	FScopp83	S
Mg2Ge	FScopp83	S
Mg2Ni	FScopp83	S
Mg2Pb	FScopp83	S
Mg2Pt	FScopp83	S
Mg2Si	FScopp83	S
Mg2Si6Al3	FScopp83	S1 S2
Mg2Sn	FScopp83	S
Mg2Zn	FScopp83	S
Mg2Zn11	FScopp83	S
Mg2Zn3	FScopp83	S
Mg3Bi2	FScopp83	S1 S2
Mg3P2	FScopp83	S
Mg3Pt	FScopp83	S
Mg3Sb2	FScopp83	S1 S2
Mg3Si6Al2	FScopp83	S1 S2
Mg4Si4Al4	FScopp83	S
Mg4Si6Al1	FScopp83	S1 S2
Mg4Si6Al2	FScopp83	S
Mg4Si7	FScopp83	S
Mg4Si8	FScopp83	S
Mg51Zn20	FScopp83	S
Mg5Ga2	FScopp83	S
Mg5In2	FScopp83	S
Mg5Si6	FScopp83	S1 S2
Mg6Pt	FScopp83	S
Mg6Si3	FScopp83	S
Mg6Si4	FScopp83	S
Mg9Si5	FScopp83	S
Mg9Si7Al3	FScopp83	S
Mg9Si7Al5	FScopp83	S
Mg9Si9Al3	FScopp83	S
MgAl2O4	FScopp83	S
MgBe13	FScopp83	S
MgCo2	FScopp83	S
MgGa	FScopp83	S
MgGa2	FScopp83	S
MgH2	FScopp83	S
MgNi2	FScopp83	S
MgO	FScopp83	S
MgPt	FScopp83	S
MgPt3	FScopp83	S
MgPt7	FScopp83	S
MgS	FScopp83	S

MgSi	FScopp83	S
MgSi ₂ Al ₂	FScopp83	S
MgSi ₆ Al ₄	FScopp83	S ₁ S ₂
MgZn ₂	FScopp83	S ₁ S ₂
Mn	FScopp83	S ₁ S ₂ S ₃ S ₄ S ₅ S ₆ L
Mn ₁₁ Si ₁₉	FScopp83	S
Mn ₁₉ Sn ₆	FScopp83	S
Mn ₂₃ C ₆	FScopp83	S
Mn ₂ Bi	FScopp83	S
Mn ₂ CaAl ₁₀	FScopp83	S
Mn ₂ P	FScopp83	S
Mn ₂ Sb	FScopp83	S ₁ S ₂
Mn ₂ Sn	FScopp83	S
Mn ₂ Ti	FScopp83	S
Mn ₂ Ti ₂	FScopp83	S
Mn ₂ Zr	FScopp83	S
Mn ₃ AlC	FScopp83	S
Mn ₃ C	FScopp83	S
Mn ₃ C ₂	FScopp83	S
Mn ₃ Ni	FScopp83	S
Mn ₃ P	FScopp83	S
Mn ₃ Si	FScopp83	S
Mn ₃ Ti	FScopp83	S
Mn ₅ C ₂	FScopp83	S
Mn ₅ Si ₃	FScopp83	S ₁ S ₂
Mn ₅ SiC	FScopp83	S
Mn ₅ Sn ₃	FScopp83	S
Mn ₆ Si	FScopp83	S
Mn ₇ C ₃	FScopp83	S
Mn ₇ Mo ₆	FScopp83	S
Mn ₈ Si ₂ C	FScopp83	S
Mn ₉ Si ₂	FScopp83	S
'Mn ₉ Ti ₂ '	FScopp83	S
MnBe ₁₂	FScopp83	S
MnBe ₂	FScopp83	S
MnBe ₅	FScopp83	S
MnBi	FScopp83	S
MnCa ₄ Al ₇	FScopp83	S
MnNi ₂	FScopp83	S
MnNi ₃	FScopp83	S
MnO	FScopp83	S
MnP	FScopp83	S
MnS	FScopp83	S ₁ S ₂
MnS ₂	FScopp83	S
MnSe ₂	FScopp83	S
MnSi	FScopp83	S
MnSn ₂	FScopp83	S
'MnTi'	FScopp83	S
Mo	FScopp83	S ₁ S ₂ S ₃ L
Mo ₂₃ C ₆	FScopp83	S
Mo ₂ C	FScopp83	S
Mo ₂ S ₃	FScopp83	S ₁ S ₂
Mo ₂ Zr	FScopp83	S
Mo ₃ Ga	FScopp83	S
Mo ₃ Ni ₁₁	FScopp83	S
Mo ₃ P	FScopp83	S ₁ S ₂
Mo ₃ Si	FScopp83	S
Mo ₄ Ni ₃	FScopp83	S
Mo ₅ Si ₃	FScopp83	S
Mo ₅ Sn ₃	FScopp83	S
MoC	FScopp83	S
MoGa ₄	FScopp83	S
MoNi ₄	FScopp83	S
MoP	FScopp83	S
MoPt	FScopp83	S

(Mn) 8.965 (Ti) 2.035

(Mn) 1.03 (Ti) 0.97

MoPt2	FScopp83	S
MoS2	FScopp83	S
MoSe2	FScopp83	S
MoSi2	FScopp83	S1 S2
MoTe2	FScopp83	S
MoZn22	FScopp83	S
MoZn7	FScopp83	S
Na	FScopp83	S1 S2 S3 L
Na2C2	FScopp83	S1 S2
Na2LiAlH6	FScopp83	S
Na2O	FScopp83	S1 S2 S3
Na2S	FScopp83	S
Na2Te	FScopp83	S
Na3AlH6	FScopp83	S1 S2
Na3As	FScopp83	S
Na3Bi	FScopp83	S
Na3Ge	FScopp83	S
Na3P	FScopp83	S
Na3Sb	FScopp83	S
NaAlH4	FScopp83	S
NaGe	FScopp83	S
NaH	FScopp83	S
NaMgH3	FScopp83	S
NaSi	FScopp83	S
NaTe	FScopp83	S
NaTe3	FScopp83	S
NaZn13	FScopp83	S
Nb	FScopp83	S L
Nb2AlC	FScopp83	S
Nb2C	FScopp83	S
Nb2Zn3	FScopp83	S
Nb3C2	FScopp83	S
Nb3P	FScopp83	S
Nb4FeSi	FScopp83	S
Nb5Si3	FScopp83	S1 S2
Nb5Sn3	FScopp83	S
Nb6Fe7	FScopp83	S
NbNi8	FScopp83	S
NbS2	FScopp83	S
NbSe2	FScopp83	S
NbSn2	FScopp83	S
NbZn	FScopp83	S
NbZn12Fe	FScopp83	S
NbZn15	FScopp83	S
NbZn2	FScopp83	S
NbZn3	FScopp83	S
NbZn7	FScopp83	S
Ni	FScopp83	S1 S2 S3 S4 S5 L
Ni10Mn3Si7	FScopp83	S
Ni11As8	FScopp83	S
Ni12P5	FScopp83	S
Ni13Ga9	FScopp83	S
Ni16Mn6Si7	FScopp83	S
Ni16Si7Ti6	FScopp83	S
Ni20Te17	FScopp83	S
Ni23C6	FScopp83	S
Ni2Cr	FScopp83	S
Ni2Ga	FScopp83	S
Ni2Ga3	FScopp83	S
Ni2Ge	FScopp83	S1 S2
Ni2In	FScopp83	S
Ni2In3	FScopp83	S
Ni2Mg3Al	FScopp83	S
Ni2Mn3Si	FScopp83	S
Ni2MnMg3	FScopp83	S

Ni2Mo	FScopp83	S
Ni2P	FScopp83	S
Ni2S	FScopp83	S
Ni2Si	FScopp83	S1 S2
Ni2Te	FScopp83	S
Ni2V	FScopp83	S
Ni3Ga	FScopp83	S
Ni3Ga4	FScopp83	S
Ni3Ga7	FScopp83	S
Ni3Ge	FScopp83	S
Ni3In	FScopp83	S
Ni3In7	FScopp83	S
Ni3Mo	FScopp83	S
Ni3P	FScopp83	S
Ni3S2	FScopp83	S
Ni3S4	FScopp83	S
Ni3Sb	FScopp83	S1 S2
Ni3Si	FScopp83	S1 S2
Ni3Si2	FScopp83	S
Ni3SiTi2	FScopp83	S
Ni3Sn	FScopp83	S1 S2
Ni3Sn2	FScopp83	S
Ni3Ti	FScopp83	S
Ni3V	FScopp83	S
Ni49Si37Ti14	FScopp83	S
Ni4Si7Ti4	FScopp83	S
Ni4W	FScopp83	S
Ni52Te40	FScopp83	S
Ni5As2	FScopp83	S
Ni5Ga3	FScopp83	S
Ni5Ge2	FScopp83	S
Ni5Ge3	FScopp83	S
Ni5P2	FScopp83	S
Ni5Sb2	FScopp83	S
Ni5Si2	FScopp83	S
'Ni5Zr4'	FScopp83	S
Ni6P5	FScopp83	S
Ni7S6	FScopp83	S
'Ni7Zr2'	FScopp83	S
'Ni8Zr3'	FScopp83	S
Ni9Mn3Si8	FScopp83	S
Ni9S8	FScopp83	S1 S2
NiAs	FScopp83	S
NiAs2	FScopp83	S1 S2
NiAss	FScopp83	S
NiFe2S4	FScopp83	S
NiGa	FScopp83	S1 S2
NiGe	FScopp83	S1 S2
NiIn	FScopp83	S
NiMnSi	FScopp83	S
NiO	FScopp83	S
NiP2	FScopp83	S
NiS	FScopp83	S1 S2
NiS2	FScopp83	S
NiSb	FScopp83	S
NiSb2	FScopp83	S
NiSe2	FScopp83	S
NiSi	FScopp83	S1 S2
NiSi2	FScopp83	S
NiSi4Ti4	FScopp83	S
NiSiTi	FScopp83	S
NiSn	FScopp83	S1 S2
NiTe	FScopp83	S
NiTe2	FScopp83	S
NiTi2	FScopp83	S1 S2

(Ni) 4.95 (Zr) 4.05

(Ni) 7.02 (Zr) 1.98
(Ni) 7.975 (Zr) 3.025

NiV3	FScopp83	S
NiW	FScopp83	S
NiW2	FScopp83	S
NiZn8	FScopp83	S
NiZr	FScopp83	S
NiZr2	FScopp83	S
P	FScopp83	S1 S2 S3 S4 L
Pb	FScopp83	S1 S2 S3 S4 S5 S6 S7 L
Pb2Au	FScopp83	S
'Pb2Pd'	FScopp83	S
'Pb2Pd3'	FScopp83	S
Pb2Zr	FScopp83	S
Pb3Au	FScopp83	S
Pb3O4	FScopp83	S
Pb3Pd5	FScopp83	S
Pb3Zr5	FScopp83	S
Pb5Li22	FScopp83	S
PbO	FScopp83	S1 S2
PbO2	FScopp83	S
PbPd	FScopp83	S
PbPd3	FScopp83	S
PbS	FScopp83	S
PbSe	FScopp83	S
PbTe	FScopp83	S
PbZr4	FScopp83	S
Pd	FScopp83	S1 S2 L
Pd2Ti	FScopp83	S
Pd3Ti	FScopp83	S
Pd3Ti2	FScopp83	S
Pd5Ti3	FScopp83	S
PdTi	FScopp83	S1 S2
PdTi2	FScopp83	S
Pt	FScopp83	S1 S2 L
Pt17Si8	FScopp83	S1 S2
Pt25Si7	FScopp83	S
Pt2Ge	FScopp83	S
Pt2Ge3	FScopp83	S
Pt2Si	FScopp83	S1 S2
Pt2Sn3	FScopp83	S
Pt2V	FScopp83	S
Pt3Ge	FScopp83	S
Pt3Ge2	FScopp83	S
Pt3Pb	FScopp83	S
Pt3Si	FScopp83	S1 S2
Pt3Sn	FScopp83	S
Pt3Ti	FScopp83	S
Pt3Ti4	FScopp83	S
Pt5Si2	FScopp83	S
Pt6Si5	FScopp83	S
Pt8Ti	FScopp83	S
PtAs2	FScopp83	S
PtGe	FScopp83	S
PtGe2	FScopp83	S
PtPb	FScopp83	S
PtPb4	FScopp83	S
PtS2	FScopp83	S1 S2
PtSb2	FScopp83	S
PtSe2	FScopp83	S1 S2
PtSi	FScopp83	S
PtSn	FScopp83	S
PtSn2	FScopp83	S
PtSn4	FScopp83	S
PtTe2	FScopp83	S1 S2
PtTi	FScopp83	S1 S2
PtTi3	FScopp83	S

(Pb) 2.001 (Pd) 0.999
(Pb) 2.05 (Pd) 2.95

S	FScopp83	S1	S2	S3	S4	S5	L
Sb	FScopp83	S1	S2	S3	L		
Sb ₂ S ₃	FScopp83	S					
Sb ₂ Se ₃	FScopp83	S					
Sb ₂ Sn ₃	FScopp83	S					
Sb ₂ SnZn	FScopp83	S					
SbFe ₂	FScopp83	S					
SbNi ₂	FScopp83	S					
Se	FScopp83	S1	S2	L			
Si	FScopp83	S1	S2	S3	S4	S5	S6
Si ₂ Te ₃	FScopp83	S1	S2				L
Si ₂ V	FScopp83	S					
Si ₂ W	FScopp83	S					
Si ₂ Zr ₃	FScopp83	S					
Si ₃ Ti	FScopp83	S					
Si ₃ Zr ₅	FScopp83	S1	S2				
Si ₄ Zr ₅	FScopp83	S					
Si ₅ V ₆	FScopp83	S					
Si ₆ A ₁₅	FScopp83	S1	S2				
SiAs	FScopp83	S					
SiAs ₂	FScopp83	S					
SiC	FScopp83	S1	S2	S3	S4	S5	S6
SiO ₂	FScopp83	S1	S2	S3	S4	S5	S6
SiP	FScopp83	S1	S2				
SiP ₂	FScopp83	S					
SiS	FScopp83	S					
SiS ₂	FScopp83	S					
SiTe ₂	FScopp83	S1	S2				
SiZr	FScopp83	S					
SiZr ₂	FScopp83	S					
Sn	FScopp83	S1	S2	S3	S4	S5	L
Sn ₁₄ Ca ₁₂ Mg ₇	FScopp83	S					
Sn ₂₀ Ca ₃₁	FScopp83	S					
Sn ₂₃ Ca ₃₆	FScopp83	S					
Sn ₂ Li ₅	FScopp83	S					
Sn ₂ Na	FScopp83	S					
Sn ₂ Zr	FScopp83	S					
Sn ₃ Ca	FScopp83	S					
Sn ₃ Ca ₅	FScopp83	S					
Sn ₃ Li ₇	FScopp83	S					
Sn ₃ Li ₈	FScopp83	S					
Sn ₃ Na	FScopp83	S					
Sn ₃ Na ₄	FScopp83	S					
Sn ₄ As ₃	FScopp83	S					
Sn ₄ Li ₁₇	FScopp83	S					
Sn ₄ Na	FScopp83	S					
Sn ₄ Na ₁₅	FScopp83	S					
Sn ₄ Na ₉	FScopp83	S					
Sn ₄ P ₃	FScopp83	S					
Sn ₅ Li ₁₃	FScopp83	S					
Sn ₅ Li ₂	FScopp83	S					
Sn ₆ Na	FScopp83	S					
SnAg ₃	FScopp83	S					
SnAs	FScopp83	S					
SnCa	FScopp83	S					
SnFe ₂	FScopp83	S					
SnLi	FScopp83	S					
SnMgLi ₄	FScopp83	S					
SnNa	FScopp83	S1	S2				
SnNa ₃	FScopp83	S					
SnNi ₂	FScopp83	S					
SnS	FScopp83	S1	S2				
SnS ₂	FScopp83	S					
SnSe	FScopp83	S1	S2				
SnSe ₂	FScopp83	S					

SnTe	FScopp83	S
SnTi2	FScopp83	S
SnZr4	FScopp83	S
Te	FScopp83	S1 S2 L
Ti	FScopp83	S1 S2 S3 S4 S5 S6 S7 L
Ti2Al	FScopp83	S
Ti2AlC	FScopp83	S
Ti2C	FScopp83	S
Ti2Cr	FScopp83	S
Ti2H	FScopp83	S
Ti2Mn	FScopp83	S
Ti2O	FScopp83	S
Ti2O3	FScopp83	S1 S2
Ti2Zn	FScopp83	S
Ti3Al	FScopp83	S
Ti3AlC	FScopp83	S
Ti3AlC2	FScopp83	S
Ti3AlSi5	FScopp83	S
Ti3O2	FScopp83	S
Ti3P	FScopp83	S
Ti3Si	FScopp83	S
Ti3Sn	FScopp83	S
Ti4AlSi7	FScopp83	S
Ti5Si3	FScopp83	S1 S2
Ti5Si4	FScopp83	S
Ti5Sn3	FScopp83	S1 S2
Ti6Sn5	FScopp83	S1 S2
TiC	FScopp83	S
TiH2	FScopp83	S
TiO	FScopp83	S
TiO2	FScopp83	S1 S2
TiSi	FScopp83	S
TiSi2	FScopp83	S
TiZn	FScopp83	S
TiZn10	FScopp83	S
TiZn15	FScopp83	S
TiZn2	FScopp83	S
TiZn3	FScopp83	S
TiZn5	FScopp83	S
Tl	FScopp83	S L
V	FScopp83	S1 S2 S3 S4 S5 L
V11Ge8	FScopp83	S
V17Ge31	FScopp83	S
V23C6	FScopp83	S
V2AlC	FScopp83	S
V2Ga5	FScopp83	S
V2Zr	FScopp83	S
V3C2	FScopp83	S
V3Ga	FScopp83	S
V3Ge	FScopp83	S
V3P	FScopp83	S
V3Si	FScopp83	S
V4Zn5	FScopp83	S
V5C2	FScopp83	S
V5Ge3	FScopp83	S
V5Si3	FScopp83	S1 S2
V6Ga5	FScopp83	S
V7Al45	FScopp83	S
V8Ga41	FScopp83	S
VSn2	FScopp83	S
VZn16	FScopp83	S
VZn3	FScopp83	S
W	FScopp83	S1 S2 S3 L
W2Zr	FScopp83	S
W3C2	FScopp83	S

Modeling

In the thermodynamic assessments, the liquid phase has been described using the Modified Quasichemical Model in the Pair Approximation (MQMPA) in order to evaluate short-range order (SRO) between the elements. Within this model, it is possible to force some binary interactions to respect a Bragg-Williams (BW) type of configurational entropy (which neglects SRO for the specific binary interactions). This makes possible the incorporation of several optimized binary systems coming from the CALPHAD community where a Bragg-Williams configurational entropy with a Redlich-Kister-Muggianu (R-K) polynomial expression of the excess Gibbs energy are typically used. For each binary system in the present database, the type of configurational entropy is given in the "List of optimized systems and calculated binary phase diagrams". In many cases, an important part of the experimental information on a given binary system is associated to solid-liquid equilibria. Hence a thermodynamic model for the liquid phase that takes into account SRO can permit a better fit of the thermodynamic and phase equilibrium data for the important solid phases that constitute the alloy.

Most solid solution phases have been described using sublattice models based on the Compound Energy Formalism (long-range order (LRO)) which include interstitials and vacancies where appropriate. The FCC-A1, BCC-A2 and HCP-A3 solid solutions have C, H and O as interstitial species.

Table 4: Optimized binary systems of the FScopp 8.3 Database (quality → Figure 1)

System	Liquid Model	Comments	Source
Ag - Al	Bragg-Williams R-K Polynomial		S.S. Lim, P.L. Rossiter, J.W. Tibbals, Calphad 19 (1995) 131-142.
Ag - As	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Ag - Au	MQMPA (Pair Fraction Exp.)	BW liquid converted to MQMPA, no ordered phases	M. Hassam, J. Agren, M. Gaune-Escard, J.P. Bros, Metall. Trans. 21A (1990) 1877-1884.
Ag - Be	Bragg-Williams R-K Polynomial		SGTE
Ag - Bi	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - C	MQMPA (Pair Fraction Exp.)	graphite saturation	P. Chartrand, CRCT, 2014
Ag - Ca	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", 2014
Ag - Cd	Bragg-Williams R-K Polynomial		Fisher et al., CALPHAD, 2019, 64, 292-305
Ag - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ag - Cr	Bragg-Williams R-K Polynomial		SGTE
Ag - Cu	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - Fe	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ag - Ga	Bragg-Williams R-K Polynomial	Stoichiometric Ag ₂ Ga	W. Gierlotka, D. Jendrzejczyk-Handzlik / Journal of Alloys and Compounds 509 (2011) 38-42
Ag - Ge	Bragg-Williams R-K Polynomial		P.Y. Chevalier, Thermochimica Acta 130 (1988) 25-32.
Ag - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ag-rich region	P. Chartrand, CRCT, 2014
Ag - Hg	Bragg-Williams R-K Polynomial		Y. Liu et al. / Thermochimica Acta 547 (2012) 83-88
Ag - In	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31.
Ag - Li	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
Ag - Mg	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31.
Ag - Mn	Bragg-Williams R-K Polynomial		I. Karakaya, W.T. Thompson, Bull. Alloy Phase Diagrams, 1990, 11, (5), 80-486
Ag - Mo	MQMPA (Pair Fraction Exp.)		Refitted from unpublished assessment of J.Korb
Ag - Na	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", <i>Journal of Chemical Thermodynamics</i> , 66 (2013) 22-33
Ag - Nb	Bragg-Williams R-K Polynomial		SGTE
Ag - Ni	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database (or SGPS) Valid for Ag-rich region	P. Chartrand, CRCT, 2021
Ag - P	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Ag - Pb	Bragg-Williams R-K Polynomial		Leo Lukas (Unpublished work, 2000) based on original work of Zimmermann (B. Zimmerman, Thesis, University of Stuttgart

			1976 "Optimisation by experimental and calculation of the binary and ternary systems of Ag, Bi, Pb and Tl".
Ag - Pd	Bragg-Williams R-K Polynomial		G. Ghosh, C. Kantner, G. B. Olson, J. Phase Equilib., 1999, 20(3), 295-308
Ag - Pt	Bragg-Williams R-K Polynomial		P.J. Spencer, 1998, based on I. Karakaya and W.T. Thompson, Bull. Alloy Phase Diagrams 8 (1987) 334.
Ag - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Ag - Sb	Bragg-Williams R-K Polynomial		E. Zoro, C. Servant, B. Legendre, Journal of Phase equilibria and Diffusion, 2007, 28, 250-257
Ag - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Ag - Si	Bragg-Williams R-K Polynomial		P.Y. Chevalier, Thermochimica Acta 113 (1988) 33-41.
Ag - Sn	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31
Ag - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Ag - Ti	Bragg-Williams R-K Polynomial		Mei Li, Changrong Li, Fuming Wang, Weijing Zhang, CALPHAD, 2005, 29, 269-275
Ag - Tl	Bragg-Williams R-K Polynomial		H. L. Lukas, unpublished reassessment based on data set collected in Zimmermanns thesis 1976 (1994).
Ag - V	Bragg-Williams R-K Polynomial		SGTE
Ag - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Ag - Zn	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
Ag - Zr	MQMPA (Pair Fraction Exp.)		Y-B. Kang & I.-H. Jung, Intermetallics 2010
Al - As	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006 (VLAB Project)
Al - Au	Bragg-Williams R-K Polynomial		J.Murray, H.Okamoto, T.B.Massalski, Bull.Alloy Phase Diags.8 (1987) 20-30 (Modified by A.T.Dinsdale to give compatibility with SGTE unary data and to prevent high-temp. stability of fcc)
Al - Be	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
Al - Bi	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Al-Bi, Al-Sb, Mg-Al-Bi and Mg-Al-Sb systems"Calphad, 2010, vol. 34, pp. 51-63.
Al - C	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2004
Al - Ca	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
Al - Cd	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2017
Al - Co	Bragg-Williams R-K Polynomial	Stoichiometric AlCo-B2	N. Dupin, I. Ansara, La Revue de Metallurgie-CIT/Sceince et Genie des Materiaux, September 1998, 1121-1129, "Thermodynamic evaluation of the system Al-Co"
Al - Cr	MQMPA (Pair Fraction Exp.)		S.Cui and I.-H.Jung, J. Alloys and Compounds, no.708, 2017 pp.887-902
Al - Cu	Bragg-Williams R-K Polynomial		COST 507, pp.28-33; Vol.Data: C. Robelin, CRCT, 2006 (VLAB Project)
Al - Fe	MQMPA (Pair Fraction Exp.)	Missing Fe ₃ Al D0 ₃ (lt)	A. T. Phan & Y-B Kang, Acta Materialia 2014
Al - Ga	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2016 modified from A.Watson CALPHAD 1992
Al - Ge	Bragg-Williams R-K Polynomial		I. Ansara, J. P. Bros, M. Gambino, CALPHAD 1979, 3(3), 225
Al - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Al - AlH₃	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Al - Hg	Bragg-Williams R-K Polynomial		A. J. McAlister, Bull. Alloy Phase Diagrams, 1985, 6, (3), 219-221

Al - In	Bragg-Williams R-K Polynomial		I. Ansara, C Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson, CALPHAD 1994, 18(4), 177-222
Al - Li	MQMPA (Pair Fraction Exp.)		J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Al - Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2006; Vol.Data: F.Gemme, CRCT, 2003 (VLAB project); C.Aliravci, 2007 (GM Project).
Al - Mn	MQMPA (Pair Fraction Exp.)		M.-S. Kim and Y-B Kang, JPED 2015
Al - Mo	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
Al - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2000 (FACT Consortium Project)
Al - Nb	Bragg-Williams R-K Polynomial		C Servant and I. Ansara, J. Chim. Phys. 1997, 94, 869-888
Al - Ni	Bragg-Williams R-K Polynomial		COST-507 K. Hack, GTT-Technologies, private communication, 2003
Al - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Al - AlO	P. Chartrand, CRCT, 2021
Al - P	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Al - Pb	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2009
Al - Pt	Bragg-Williams R-K Polynomial		K.Wu and Z.Jin, Journal of Phase Equilibria, 21(3), 2000, 221-226
Al - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2007; based on R.C.Sharma and Y.A.Chang BAPD, 8(2) 1987 p.128-131
Al - Sb	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Al-Bi, Al-Sb, Mg-Al-Bi and Mg-Al-Sb systems"Calphad, 2010, vol. 34, pp. 51-63.
Al - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Al - Si	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006; Vol. Data: F. Gemme, CRCT, 2003 (VLAB project)
Al - Sn	MQMPA (Pair Fraction Exp.)		Y.-B. Kang and A.D. Pelton, CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 180 188
Al - Ti	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
Al - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018; from McAlister 1989 BAPD
Al - V	Bragg-Williams R-K Polynomial		COST 507, pp.95-98.
Al - W	Bragg-Williams R-K Polynomial		COST 507, pp.103-108.
Al - Zn	MQMPA (Pair Fraction Exp.)		Liquid: P. Chartrand, CRCT, 2006 (VLAB Project); Vol.Data. F.Gemme, CRCT, 2003 (VLAB Project); solids S. -L. Chen and Y.A. Chang, Calphad, 17 (1993), pp. 113-124.
Al - Zr	Bragg-Williams R-K Polynomial		COST 507, pp.112-116.
As - Au	Bragg-Williams R-K Polynomial		P.J.Spencer, June 1998
As - Bi	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
As - Cd	MQMPA (Pair Fraction Exp.)		O. Kidari, Master Thesis, CRCT, 2023
As - Cu	Bragg-Williams R-K Polynomial	Stoichiometric Cu ₃ As	M.Hamalainen, private communication
As - Fe	Bragg-Williams R-K Polynomial	Missing stoichiometric compounds	P. J. Spencer, 2008
As - Ga	Bragg-Williams R-K Polynomial		C Chatillon, I. Ansara, A. Watson and B. B. Argent: CALPHAD, 1990, 14(2), 203-14.
As - Ge	Bragg-Williams R-K Polynomial		I. Ansara and D. Dutarte: CALPHAD, 1984, 8(4), 323-342
As - In	Bragg-Williams R-K Polynomial		C.Chatillon, I.Ansara, A.Watson, B.B.Argent, Calphad 14 (1990) 203-214
As - Ni	Bragg-Williams R-K Polynomial		S. Uhland et al., CALPHAD 25[1] (2001) 109-124
As - P	Bragg-Williams R-K Polynomial		I Ansara, C Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD 1994, 18(4), 177-222

As - Pb	Bragg-Williams R-K Polynomial		M.Hamalainen, private communication
As - Pt	Bragg-Williams R-K Polynomial		M. Li et al. / Journal of Alloys and Compounds 437 (2007) 71–79
As - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2021; from Prostakova, Shishin, Jak 2021 (change in g0 sulfur)
As - Sb	Bragg-Williams R-K Polynomial		H.Ohtani, Calphad 18 (1994) 196
As - Si	MQMPA (Pair Fraction Exp.)		P.-J. Spencer, 2006
As - Sn	Bragg-Williams R-K Polynomial		L.Wu et al., CALPHAD (74) 2021, 102296
As - Te	MQMPA (Pair Fraction Exp.)	Approx..	P. Chartrand, CRCT, 2018
As - Zn	MQMPA (Pair Fraction Exp.)		O. Kidari, Master Thesis, CRCT, 2023
Au - Bi	Bragg-Williams R-K Polynomial		C. Servant, E. Zoro and B. Legendre, Calphad, 30 (2006) 443
Au - C	Bragg-Williams R-K Polynomial		P.J. Spencer (2007)
Au - Co	Bragg-Williams R-K Polynomial		H.Q.Dong et al., J. Electronic Mater., 38(10) 2009, p.2158
Au - Cr	Bragg-Williams R-K Polynomial		P.J. Spencer, (1998)
Au - Cu	Bragg-Williams R-K Polynomial	No ordered L1 ₂ & L1 ₀ phases	B.Sundman, S.G.Fries, A.Oates, Calphad 22 (1998) 335-354
Au - Fe	Bragg-Williams R-K Polynomial		Y.Liu et al., CALPHAD 476 (2009) 79-83
Au - Ga	Bragg-Williams R-K Polynomial		J. Wang, Y. Liu, L. Liu, H. Zhou, Z. Jin, Calphad 35(2) (2011) 242-248
Au - Ge	Bragg-Williams R-K Polynomial		P. Y. Chevalier: Thermochimica Acta, 1989, 141, 217-22
Au - In	Bragg-Williams R-K Polynomial		I.Ansara, J.P.Nabot, Calphad 16 (1992) 13-18
Au - Ni	Bragg-Williams R-K Polynomial		J. Wang et al., CALPHAD (2005) pp.263-268
Au - Pb	Bragg-Williams R-K Polynomial		J.P.Nabot, Thesis, Grenoble,1986
Au - Pd	Bragg-Williams R-K Polynomial		COST-531
Au - Pt	Bragg-Williams R-K Polynomial		P.J. Spencer, 1994
Au - Sb	Bragg-Williams R-K Polynomial		E.Zoro, C.Servant and B.Legendre, J.Thermal Anal. & Calo., 90[2] (2007) pp.347-353
Au - Si	Bragg-Williams R-K Polynomial		SGTE
Au - Sn	Bragg-Williams R-K Polynomial		P.Y.Chevalier, Thermochimica Acta 130 (1988) 1-13
Au - Te	Bragg-Williams R-K Polynomial		Y. Feutelais, D. Mounai, J. R. Didry, B. Legendre: J. Phase Equil., 1994, 15(4), 380-385 (modified by SGTE)
Au - Ti	Bragg-Williams R-K Polynomial		W.Luo et al., CALPHAD 25(1) 2001 19-26
Au - Tl	Bragg-Williams R-K Polynomial		P.Y. Chevalier: Thermochimica Acta, 1989, 155, 211-225
Au - Zn	Bragg-Williams R-K Polynomial	Intermetallic solutions missing – liquid and FCC-A1 only	COST-531
Be - Ca	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Cu	Bragg-Williams R-K Polynomial		P.J. Spencer, 2003
Be - Fe	Bragg-Williams R-K Polynomial	Stoichiometric BeFe	M. Piché, M.A.Sc., École Polytechnique, 2002
Be - Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Mg	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
Be - Mn	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
Be - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Si	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Sn	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Cd	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

Bi - Cr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Cu	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Bi - Fe	Bragg-Williams R-K Polynomial		from FSstel
Bi - Ga	MQMPA (Pair Fraction Exp.)		B.Kumar et al., CALPHAD 74 (2021)
Bi - Ge	Bragg-Williams R-K Polynomial		P. Y. Chevalier: Thermochimica Acta, 1988, 132, 111-116
Bi - Hg	Bragg-Williams R-K Polynomial		Unpublished assessment from S. A. Mucklejohn (SGTE)
Bi - In	Bragg-Williams R-K Polynomial		D.Boa, I.Ansera, Thermochimica Acta 314 (1998) 79-86
Bi - Li	MQMPA (Pair Fraction Exp.)		Z.M. CAO, W. XIE, P. CHARTRAND, S.H. WEI, G.W. DU and Z.-Y. QIAO, "Thermodynamic assessment of the Bi-alkali metal (Li, Na, K, Rb) systems using the modified quasichemical model for the liquid phase", CALPHAD, 46 (2014), pp.159-167
Bi - Mg	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Mg-Bi and Mg-Sb systems and short-range-ordering behavior of the liquid solutions" Calphad, 2009, vol. 33, pp. 744-754.
Bi - Na	MQMPA (Pair Fraction Exp.)		Z.M. CAO, W. XIE, P. CHARTRAND, S.H. WEI, G.W. DU and Z.-Y. QIAO, "Thermodynamic assessment of the Bi-alkali metal (Li, Na, K, Rb) systems using the modified quasichemical model for the liquid phase", CALPHAD, 46 (2014), pp.159-167
Bi - Ni	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Bi - Pb	Bragg-Williams R-K Polynomial		D.Boa, I.Ansera, Thermochimica Acta 314 (1998) 79-86
Bi - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Bi - Sb	MQMPA (Pair Fraction Exp.)		P. Chartrand 2018 modified from P.J. Spencer, 2006
Bi - Si	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
Bi - Sn	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Bi - Te	MQMPA (Pair Fraction Exp.)		B.Kumar et al. (Paliwal) CALPHAD 74 (2021)
Bi - Ti	Bragg-Williams R-K Polynomial		unpublished assessment of H.L. Lukas based on an earlier assessment of Zimmermann B., Henig E. T., Lukas H. L.: Z. Metallkde., 1976, 67(12), 815-820.
Bi - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Bi - Zn	Bragg-Williams R-K Polynomial		C.Girard, Thesis, Marseille, 1985
C - Ca	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
C - Co	Bragg-Williams R-K Polynomial		A. Fernandez Guillermet: Z. Metallkde., 1987, 78, 700-9
C - Cr	Bragg-Williams R-K Polynomial		B. J. Lee: CALPHAD 1992, 16(2), 121-149
C - Cu	MQMPA (Pair Fraction Exp.)		S. Shubhank and Y.-B.Kang CALPHAD 45 (2014) 127–137
C - Fe	MQMPA (Pair Fraction Exp.)	Incl. metastable carbides	M-S Kim, Y-B Kang JPE 2015
C - Ge	MQMPA (Pair Fraction Exp.)		P.Chartrand, 2014
C - Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
C - Mn	MQMPA (Pair Fraction Exp.)		M.K. Paek, Y-B Kang, CALPHAD 46 (2014), 92-102
C - Mo	Bragg-Williams R-K Polynomial		SGTE
C - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2013
C - Nb	Bragg-Williams R-K Polynomial		WM Huang: Mater. Sci. and Techn. 1990, 6(8), 687-694
C - Ni	Bragg-Williams R-K Polynomial		B. J. Lee: CALPHAD, 1992, 16(2), 121-149
C - P	Bragg-Williams R-K Polynomial		P. Gustafson: Inst. Met. Res. (Report IM-2549, 1990))
C - Pb	Bragg-Williams R-K Polynomial		unpublished assessment of T. G. Chart, NPL 1987
C - Pt	Bragg-Williams R-K Polynomial		GTT, 2005
C - S	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FACTPS Database	Fahmi TAFWIDL and Youn-Bae KANG, ISIJ, 2017

C - Sb	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
C - Si	MQMPA (Pair Fraction Exp.)		M-K Paek, Y-B Kang CALPHAD 46 (2014) 92–102
C - Sn	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2007
C - Ti	MQMPA (Pair Fraction Exp.)	(C solubility in Ti-HCP is neglected)	P.J. Spencer, 2007 (VLAB Project)
C - V	Bragg-Williams R-K Polynomial		WM Huang : Z. Metallkde, 1991, 82, (3), 174-181
C - W	Bragg-Williams R-K Polynomial		P. Gustafson: Report TRITA 0212 (1985), Mat. Sci and Tech. 1986, 2(7), 653-658
C - Zr	Bragg-Williams R-K Polynomial		A. Fernandez Guillermet: J. Alloys Compounds, 1995, 217, 69-89
Ca - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ca - Cr	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ca - Cu	MQMPA (Pair Fraction Exp.)		Jian Wang et al., Calphad 75 (2021) 102325
Ca - Fe	MQMPA (Pair Fraction Exp.)		S.Cui, M.Paliwal and I.-H. Jung, MetTrans 2014
Ca - Ga	Bragg-Williams R-K Polynomial		X.Li et al., CALPHAD 43 (2013) 52-60
Ca - Ge	MQMPA (Pair Fraction Exp.)		P.Chartrand, CRCT, 2023
Ca - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ca - CaH ₂	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Ca - In	MQMPA (Pair Fraction Exp.) + BW		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
Ca - Li	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
Ca - Mg	MQMPA (Pair Fraction Exp.)		J. Wang, Ph.D. Thesis, Ecole Polytechnique, 2014
Ca - Mo	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ca - Mn	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006 (GM project)
Ca - Na	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", Journal of Chemical Thermodynamics, 66 (2013) 22-33
Ca - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
Ca - Ni	Bragg-Williams R-K Polynomial		M. Medraj, 2006
Ca - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Ca - CaO	P. Chartrand, 2000
Ca - Pb	Bragg-Williams R-K Polynomial		V.P.Itkin and C.B.Alcock, J. Phase Equilib. 1992, pp.162-169
Ca - Si	MQMPA (Pair Fraction Exp.)		M. HEYRMAN and P. CHARTRAND, "Thermodynamic Evaluation and Optimization of the Ca-Si System", <i>J. Phase Equilibria and Diffusion</i> , 27 [3], 220-230, 2006
Ca - Sn	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
Ca - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ca - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ca - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ca - Zn	MQMPA (Pair Fraction Exp.)		P. Spencer, A.D. Pelton., Y.-B. Kang, P. Chartrand, and C. Fuerst, Calphad 32 (2007), pp. 423-431
Ca - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Cd - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd - Cr	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018

Cd – Cu	Bragg-Williams R-K Polynomial		X-M Chen, L-B Liu, L-G Zhang, H. Bo and Z-P Jin Trans Nonferrous Met Soc. China, 20 (2010) 649-654
Cd – Fe	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – Ga	Bragg-Williams R-K Polynomial		Zakulski W., Moser Z., Rzyman K., Lukas H. L., Fries S. G., Sikiennik M., Kaczmarczyk R., Castanet R.: J. Phase Equil., 1993, 14(2), 184-196.
Cd – Ge	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – Hg	Bragg-Williams R-K Polynomial		Jang J., Silk N. J., Watson A., Bryant A. W., Chart T. G., Argent B. B.: CALPHAD, 1995, 19(3), 415-430
Cd – In	Bragg-Williams R-K Polynomial		Zakulski W., Moser Z., Rzyman K., Lukas H. L., Fries S. G., Sikiennik M., Kaczmarczyk R., Castanet R.: J. Phase Equil. 1993, 14(2), 184-196
Cd – Mg	Bragg-Williams R-K Polynomial		Ren, Li, Guo, Du, Thermochimica Acta 2012
Cd – Mn	Bragg-Williams R-K Polynomial		Qin et al, CALPHAD 47 (2014) 83-91
Cd – Na	Bragg-Williams R-K Polynomial		Y. Liu et al. / Journal of Alloys and Compounds 473 (2009) 60–64
Cd – Ni	Bragg-Williams R-K Polynomial		H.Azza, N.Selhaoui, L.Bouirden, Int. J. Adv. Res. in Phys. Sc. 5[5] (2018) pp.17-23
Cd – Pb	Bragg-Williams R-K Polynomial		W. Zakulski, Z. Moser: J. Phase Equilib, 1995, 16(3), 239-242. W. Zakulski, Z. Moser: J. Phase Equilib, 1995, 16(6), 484
Cd – S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Cd – Sb	Bragg-Williams R-K Polynomial		L. A. Zabdyr: CALPHAD 1997, 21(3), 349-358.
Cd – Si	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – Sn	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – Ti	Bragg-Williams R-K Polynomial		C.Zhang et al., CALPHAD 42 (2013) 6-12
Cd – Tl	Bragg-Williams R-K Polynomial		Y. Liu et al. / Journal of Alloys and Compounds 473 (2009) 60–64
Cd – V	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – W	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cd – Zn	Bragg-Williams R-K Polynomial		L. A. Zabdyr: CALPHAD 1997, 21(3), 349-358
Co – Cr	MQMPA (Pair Fraction Exp.)		A.Kusoffsky, B.Jansson, Calphad 21 (1997) 321-333; liquid converted to MQMPA [2022Cha]
Co – Cu	Bragg-Williams R-K Polynomial		P.J.Spencer - L.Kaufman interaction parameters with SGTE element data
Co – Fe	MQMPA (Pair Fraction Exp.)		A.F.Guillermet, High Temp. High Press. 19 (1988) 477-499; liquid converted to MQMPA [2021Cha]
Co – Ga	Bragg-Williams R-K Polynomial		A. Chari et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 189–195
Co – In	Bragg-Williams R-K Polynomial		D. Boa, B. K. Dongui, I. Ansara: CALPHAD 25 (2001) 645-650
Co – Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Co – Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Co – Mn	Bragg-Williams R-K Polynomial		W.Huang, Calphad 13 (1989) 231-242
Co – Mo	Bragg-Williams R-K Polynomial		SGTE
Co – Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Co – Nb	Bragg-Williams R-K Polynomial		K.C.H.Kumar, I.AnSara, P.Wollants, L.Delaey, J.Alloys and Cpds. 267 (1998) 105-112
Co – Ni	Bragg-Williams R-K Polynomial		A.F.Guillermet, Z.Metallkde. 78 (1987) 639-647; Z.Metallkde. 79 (1988) 524-536
Co – O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Co – CoO	P. Chartrand, CRCT, 2021
Co – Pb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Co – Pd	Bragg-Williams R-K Polynomial		G. Ghosh, C. Kantner, G.B. Olson, J.Phase Equilibria 20 (1999) 295-308

Co - Pt	Bragg-Williams R-K Polynomial		D.E. Kim et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 35 (2011) 323–330
Co - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	F. Kongoli, Y. Desseract and A.D. Pelton, "Thermodynamic Modeling of Liquid Fe-Ni-Co-Cu-S Mattes", Met. Trans.,29B, 591-601 (1998).
Co - Sb	Bragg-Williams R-K Polynomial		Y.Zhang et al., CALPHAD 33 (2009) 405–414
Co - Si	Bragg-Williams R-K Polynomial	Stoichiometric CoSi Inverted L-L miscibility gap above 3000°C	S.D.Chi, Calphad 16 (1992) 151-159
Co - Sn	Bragg-Williams R-K Polynomial		L.Liu et al., J.Electron.Mater.33(9) (2004) 935-939
Co - Te	Bragg-Williams R-K Polynomial		H. Yuan, J. Wang, B. Hu, R. Zhao, Y. Du and S.-Y. Zhang, CALPHAD 68 (2020) 10174
Co - Ti	Bragg-Williams R-K Polynomial		G.Cacciamani, R.Ferro, I.Anbara, N.Dupin, submitted to "Intermetallics", 1999
Co - V	Bragg-Williams R-K Polynomial	Stoichiometric Co ₃ V	J. Bratberg, B. Sundman: J. Phase Equil., (2003), 24(6), 495-503
Co - W	Bragg-Williams R-K Polynomial		Markstrom, Sundman & Frisk JPE (2005)
Co - Zn	Bragg-Williams R-K Polynomial	Missing CoZn ₇ (ht) and CoZn(ht)	G.P. Vassilev, M. Jiang: J. Phase Equil. and Diffusion 2004, 25, 259-268
Co - Zr	Bragg-Williams R-K Polynomial		Durga & Kumar, CALPHAD 2010
Cr - Cu	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung, CALPHAD 2017
Cr - Fe	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung, CALPHAD 2017
Cr - Ga	Bragg-Williams R-K Polynomial		P.Chartrand, CRCT, 2022
Cr - H	MQMPA (Pair Fraction Exp.)	estimated from Ti-H; Gaseous species must be taken from FactPS Database Valid for Cr rich region	P. Chartrand 2010
Cr - Hg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cr - Li	MQMPA (Pair Fraction Exp.)	estimated from Fe-Li	P. Chartrand 2003
Cr - Mg	MQMPA (Pair Fraction Exp.)		S. Cui et al., Journal of Alloys and Compounds, 2017
Cr - Mn	Bragg-Williams R-K Polynomial		COST 507, pp.145-148.
Cr - Mo	Bragg-Williams R-K Polynomial		COST 507
Cr - Na	MQMPA (Pair Fraction Exp.)	estimated from Fe-Na	P. Chartrand, CRCT, 2003
Cr - Nb	Bragg-Williams R-K Polynomial		J.G.Costa Neto, S.G.Fries, H.-L.Lukas, Calphad 17 (1993) 219-228
Cr - Ni	Bragg-Williams R-K Polynomial		B.J. Lee, Calphad 16 1992, pp. 121-149.
Cr - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Cr - CrO	P.Chartrand, CRCT, 2021
Cr - P	Bragg-Williams R-K Polynomial	Valid in the Cr - Cr ₂ P region only	SGTE
Cr - Pb	Bragg-Williams R-K Polynomial		P. Chartrand, 2003
Cr - Pt	Bragg-Williams R-K Polynomial		Oikawa et al. Journal of Magnetism and Magnetic Materials 236 (2001) 220–233
Cr - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P.Waldner and W.Sitte, Int. J. Mat. Res. (formerly Z. Metallkd.) 102 (2011) 10, 1216-1225.
Cr - Si	MQMPA (Pair Fraction Exp.)		Senlin Cui & In-Ho Jung, Met.Trans.2017
Cr - Sn	Bragg-Williams R-K Polynomial		R. Jerlerud Perez, B. Sundman: CALPHAD, 25 (2001) 59-66
Cr - Ti	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
Cr - V	Bragg-Williams R-K Polynomial		B.J.Lee, Z.Metallkde 83 (1992) 292-299
Cr - W	Bragg-Williams R-K Polynomial		P. Gustafson, Calphad 12 (1988), pp. 277-292.
Cr - Zn	Bragg-Williams R-K Polynomial		I.Anbara, COST 507 (1998) ISBN 92-828-3902-8, p.158-160

Cr - Zr	Bragg-Williams R-K Polynomial		K.Zeng, M.Hamalainen, I.Anasra, COST 507 (1998) ISBN 92-828-3902-8 p 161-164
Cu - Fe	MQMPA (Pair Fraction Exp.)		Shubhank and Y-B. Kang, CALPHAD 2014
Cu - Ga	Bragg-Williams R-K Polynomial	Missing Cu ₉ Ga ₄ γ ₁ , γ ₂ , γ ₃ solid solutions	Li et al., CALPHAD 32(2) (2008), 447-453
Cu - Ge	Bragg-Williams R-K Polynomial		Wang et al. J.Alloys Cmpds 2010
Cu - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Cu - CuH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Cu - Hg	Bragg-Williams R-K Polynomial		Y. Liu et al. / Thermochimica Acta 547 (2012) 83-88
Cu - In	Bragg-Williams R-K Polynomial	γ-CuIn, η-CuIn are missing	C.R.Kao, A.Bolcavage et al, J Phase Equilibria 14 (1993) 22-30
Cu - Li	Bragg-Williams R-K Polynomial		COST 507, pp.168-169.
Cu - Mg	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, 2017
Cu - Mn	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, 2017
Cu - Mo	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Cu - Nb	Bragg-Williams R-K Polynomial		B.J. Lee database, private communication to SGTE, 1999.
Cu - Ni	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Cu - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Cu - Cu ₂ O	P. Chartrand, CRCT, 2018
Cu - P	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Pb	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Pd	Bragg-Williams R-K Polynomial	Liquid only (Cu-rich)	G.K.Sigworth, J.F.Elliott, Canad.Met.Quarterly 13 (1974) 455-461.
Cu - Pt	Bragg-Williams R-K Polynomial		T.Abe, B.Sundman, H.Onodera, JPEDAV (2006) 27:5-13
Cu - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P.Waldner, Met.Trans.B, 51B (2020) 805-817
Cu - Sb	Bragg-Williams R-K Polynomial		SGTE 2004
Cu - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Si	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, CALPHAD, 2017
Cu - Sn	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Cu - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Ti	Bragg-Williams R-K Polynomial		K.C.H. Kumar, I. Ansara, P. Wollants, L. Delaey, Z. Metallkde. 87 (1996), pp. 666-672.
Cu - Tl	Bragg-Williams R-K Polynomial		P.Y.Chevalier, Thermochimica Acta 156 (1989) 383-392
Cu - V	Bragg-Williams R-K Polynomial		B.J. Lee database, private communication to SGTE, 1999.
Cu - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Cu - Zn	Bragg-Williams R-K Polynomial		Liang, Hsiao, Schmid-Feitzer CALPHAD, 2015
Cu - Zr	MQMPA (Pair Fraction Exp.)		D.H.Kang I.H.Jung Intermetallics 2010
Fe - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Fe-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Fe - Hg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Fe - In	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022, Modified from M.Ohno and K.Yoh, Materials Transactions, Vol. 50, No. 5 (2009) pp. 1202 to 1207
Fe - Li	MQMPA (Pair Fraction Exp.)	from Fe in liquid Li	P. Chartrand, 2003
Fe - Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2006

Fe - Mn	MQMPA (Pair Fraction Exp.)		M-S Kim, Y-B Kang, JPE, 2015
Fe - Mo	Bragg-Williams R-K Polynomial		SGTE
Fe - Na	MQMPA (Pair Fraction Exp.)	from Fe in liquid Na	P. Chartrand, CRCT, 2003
Fe - Nb	Bragg-Williams R-K Polynomial		Khvan & Hallstedt CALPHAD 2013
Fe - Ni	Bragg-Williams R-K Polynomial		A.Dinsdale, T.Chart, NPL, unpublished work, 1986: I.Ansara - fcc ordering
Fe - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Fe – FeO Missing non-stoichiometric wustite-FeO (considered stoichiometric)	P. Chartrand, CRCT, 2021
Fe - P	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Fe - Pb	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Fe - Pd	Bragg-Williams R-K Polynomial		G. Ghosh, C. Kantner, G.B. Olson, J.Phase Equilibria 20 (1999) 295-308
Fe - Pt	Bragg-Williams R-K Polynomial		P.Fredriksson & B.Sundman, CALPHAD 25(4) 2001, 535-548
Fe - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P.Waldner & A.D.Pelton, JPE, 26(1) 2005
Fe - Sb	Bragg-Williams R-K Polynomial		C.Li et al., CALPHAD 47 (2014) 23-34
Fe - Si	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung CALPHAD 2017
Fe - Sn	Bragg-Williams R-K Polynomial		K.C.H.Kumar, P.Wollants, L.Delaey, Calphad 20 (1996) 139-149 (with modifs from P. Chartrand (2018))
Fe - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Fe - Ti	Bragg-Williams R-K Polynomial		P.J. Spencer, 2001 (FACT Consortium)
Fe - Tl	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Fe - V	Bragg-Williams R-K Polynomial		W.Huang, Z.Metallkde 82 (1991) 391-401
Fe - W	Bragg-Williams R-K Polynomial		J-O Andersson and P Gustafson: CALPHAD, 1983, 7(4), 317-326
Fe - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand (2018) MQMPA calibrated on the solids of Xiong, Du Liu CALPHAD 2009 (later modifications by M.Jacobs)
Fe - Zr	MQMPA (Pair Fraction Exp.)		J.J. Marin Bejarano, CRCT, 2009
Ga - Ge	Bragg-Williams R-K Polynomial		I Ansara, J P Bros, M Gambino: CAPHAD, 1979, 3, 225-233
Ga - Hg	Bragg-Williams R-K Polynomial		unpublished assessment of I. Ansara, (1991); from SGTE
Ga - In	Bragg-Williams R-K Polynomial		B. C. Rugg, T. G. Chart: CALPHAD, 1990, 14(2), 115-123
Ga - Li	Bragg-Williams R-K Polynomial		H.Azza et al., J. Phase Equil. Diffus. (2017) 38:788–795
Ga - Mg	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CALPHAD 2014
Ga - Mo	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
Ga - Na	Bragg-Williams R-K Polynomial		J. Wang et al. / Journal of Crystal Growth 307 (2007) 59-65
Ga - Ni	Bragg-Williams R-K Polynomial		Z.-M.Cao, X.Shi, W.Xie, I.Ohnuma, K.Ishida & Z.-Y.Qiao, Rare Met. (2015) 34(2) 864-872
Ga - P	Bragg-Williams R-K Polynomial		I. Ansara, C. Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD, 1994, 18(4), 177-222
Ga - Pb	Bragg-Williams R-K Polynomial		I. Ansara, F. Ajersch: J. Phase Equil., 1991, 12(1), 73-77
Ga - Pt	Bragg-Williams R-K Polynomial		M. Li et al. / Intermetallics 14 (2006) 826–831
Ga - Sb	Bragg-Williams R-K Polynomial		I. Ansara, C. Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD, 1994, 18(4), 177-222
Ga - Si	Bragg-Williams R-K Polynomial		Olesinski BAPD (6) 1985, 362-364

Ga - Sn	MQMPA (Pair Fraction Exp.)		B.Kumar et al., Metals (2021)
Ga - Te	MQMPA (Pair Fraction Exp.)		B.Kumar et al. (Paliwal) CALPHAD 74 (2021)
Ga - Ti	Bragg-Williams R-K Polynomial		Y.Liu et al., CALPHAD 41 (2013) 140-149
Ga - Tl	Bragg-Williams R-K Polynomial		I. Katayama et al., T. Iida, Z. Metallknd. 94, 2003, p.1296
Ga - V	Bragg-Williams R-K Polynomial		D. Ling et al./CALPHAD 51(2015) 125-132
Ga - Zn	MQMPA (Pair Fraction Exp.)		Dutkiewicz, J., Moser, Z., Zabdyr, L., Gohil ,D. D., Chart, T. G., Ansara I., Girard, C.: Bull. Alloy Phase Diagrams, 1990, 11(1), 77-82; liquid converted to MQMPA [2022Cha]
Ga - Zr	Bragg-Williams R-K Polynomial		W. Luo et al. / Journal of Alloys and Compounds 587 (2014) 497–505
Ge - In	Bragg-Williams R-K Polynomial		P. Y. Chevalier: 1989, 155, 227-240
Ge - Mg	MQMPA (Pair Fraction Exp.)		Dmitri Nasyrov and In-Ho Jung, "Thermodynamic modeling of the Mg-Ge-Pb system"Calphad, 2009, vol. 33, pp. 521-529.
Ge - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023 using solids from Y. Wang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 37 (2012) 72–76
Ge - Ni	Bragg-Williams R-K Polynomial		S.Jin et al./CALPHAD 38 (2012) 23–34
Ge - P	Bragg-Williams R-K Polynomial		H. Dong, X. Tao and M. Paulasto-Krockel, Journal of Electronic Materials (2021) 50:4272-4
Ge - Pb	Bragg-Williams R-K Polynomial		P. Chevalier, Thermochemica Acta, 1989, vol 155, pp. 227-240
Ge - Pt	Bragg-Williams R-K Polynomial		J.S. Wang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 33 (2009) 561-569
Ge - Sb	Bragg-Williams R-K Polynomial		P. Y. Chevalier, Thermochemica Acta, 1989, 155, 227-240
Ge - Si	Bragg-Williams R-K Polynomial		Z. H. Long, H.S. Liu, Z.P. Jin: J. Alloys Compounds, 479 (2009) 262-267
Ge - Sn	Bragg-Williams R-K Polynomial		Y. Feutelais, B. Legendre, S. G. Fries: CALPHAD, 1996, 20(1), 109-123
Ge - Ti	Bragg-Williams R-K Polynomial		P. Chevalier, Thermochemica Acta, 1989, vol 155, pp. 227-240
Ge - V	Bragg-Williams R-K Polynomial		C.P. Wang et al. / Intermetallics 16 (2008) 544-549
Ge - Zn	Bragg-Williams R-K Polynomial		P. Chevalier, Thermochemica Acta, 1989, vol 155, pp. 227-240
H - Li	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Li - LiH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Mg	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Mg – MgH ₂	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Mn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Mn-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Na	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Na - NaH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Ni	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ni-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Si	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Si-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Sn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Si-rich region	P. Chartrand, CRCT, 2023 using model from M.-C. Heuzey, Master Thesis, CRCT

H - Ti	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ti – TiH ₂	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Zn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Zn-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Zr	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Zr-rich region	P. Chartrand, CRCT, 2014 from Ti-H of J.P. Harvey 2006
Hg – Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Hg – Pb	Bragg-Williams R-K Polynomial		A. Maitre, J. M. Fiorani, M. Vilasi: <i>J. Phase Equilib.</i> , 2002, 23(4), 329
Hg – S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Hg – Sn	Bragg-Williams R-K Polynomial		Yee-Wen Yen, Joachim Grobner, Steve C. Hansen, and Rainer Schmid-Fetzer <i>JPE</i> , 24(2), p.151-167, 2003
Hg – Zn	Bragg-Williams R-K Polynomial		S. C. Hansen: <i>CALPHAD</i> , 1998, 22, 359-373.
In - Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
In - Mg	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95.
In - Na	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", <i>Journal of Chemical Thermodynamics</i> , 66 (2013) 22-33
In - Ni	Bragg-Williams R-K Polynomial		P.Waldner and H.Ipser, <i>Z.Metallk</i> , 93(8) 2002, 825-832
In - P	Bragg-Williams R-K Polynomial		I.Ansara, C.Chatillon, <i>Calphad</i> 18 (1994) 204
In - Pb	Bragg-Williams R-K Polynomial		D.Boa, I.Ansara, <i>Thermochimica Acta</i> 314 (1998) 79-86
In - Pt	Bragg-Williams R-K Polynomial		Y. Liang et al. / <i>Journal of Alloys and Compounds</i> 475 (2009) 220–226
In - Sb	Bragg-Williams R-K Polynomial		T.J.Anderson, <i>Calphad</i> 18 (1994) 206
In - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023, refitted liquid using solids from J.-B.Li, M.-C.Record,J.-C.Tedenac, <i>Z.Metallkd</i> 94 (2003) 4, 381-389
In - Si	Bragg-Williams R-K Polynomial		R.W.Olesinski, N.Kanani, G.J.Abbaschian, <i>Bull.Alloy Phase Diags.6</i> (1985) 128-130
In - Sn	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95
In - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
In - Zn;	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95
Li - Mg	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
Li - Mn	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
Li - Mo	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2003
Li - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2001
Li - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
Li - Ni	MQMPA (Equivalent Fraction Exp.)		P. Chartrand, 2003
Li - O	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database	P. Chartrand, CRCT, 2021

Li - Pb	MQMPA (Pair Fraction Exp.)	Liquid only	P. Chartrand, 2003
Li - Sb	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Li - Si	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, CRCT, 2007
Li - Sn	MQMPA (Pair Fraction Exp.)		J. WANG, J. HAN, I.-H. JUNG, D. BAIROS and P. CHARTRAND, "Thermodynamic optimizations on the binary Li-Sn system and ternary Mg-Sn-Li system", CALPHAD, accepted, (July 2014).
Li - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Li - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Li - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022
Li - Zn	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
Li - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Mg - Mn	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton, P. Chartrand, P. Spencer and C. Fuerst, J. Phase Equil. Diff. 28 (2007), pp. 342-354
Mg - Mo	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Mg - Na	MQMPA (Equivalent Fraction Exp.)		P. Chartrand, CRCT, 2003 (FACT Consortium)
Mg - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Mg - Ni	Bragg-Williams R-K Polynomial		M.Jacobs, COST 507 report (1998) ISBN 92-828-3902-8, p.218-220
Mg - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Mg - MgO	P. Chartrand, CRCT, 2021
Mg - Pb	MQMPA (Pair Fraction Exp.)		Dmitri Nassyrov and In-Ho Jung, "Thermodynamic modeling of the Mg-Ge-Pb system"Calphad, 2009, vol. 33, pp. 521-529.
Mg - Pt	Bragg-Williams R-K Polynomial		W.Gierlotka et al., J.Mater.Res.,37(11) 2022,p.1904
Mg - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2021
Mg - Sb	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Mg-Bi and Mg-Sb systems and short-range-ordering behavior of the liquid solutions"Calphad, 2009, vol. 33, pp. 744-754
Mg - Si	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006 (VLAB Project); Vol. Data F. Gemme, CRCT, 2003 (VLAB project)
Mg - Sn	MQMPA (Pair Fraction Exp.)		Y.-B. Kang and A.D. Pelton, CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 180 188
Mg - Ti	MQMPA (Pair Fraction Exp.)		X.-F. Sheng, Projet de fin d'études, CRCT, 2008
Mg - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Mg - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Mg - Zn	MQMPA (Pair Fraction Exp.)		Liquid: P.J. Spencer, 2006; Solids from COST-507; Vol. Data liquid: F. Gemme, CRCT, 2003; Vol. Data solids C.Aliravci, CRCT, 2007
Mg - Zr	MQMPA (Pair Fraction Exp.)		J.J. Marin Bejarano, CRCT, 2009
Mn - Mo	Bragg-Williams R-K Polynomial		COST-507
Mn - Na	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
Mn - Nb	Bragg-Williams R-K Polynomial		Z.-K. Liu & B. Hallstedt CALPHAD 2012
Mn - Ni	Bragg-Williams R-K Polynomial		NPL (National Physics Laboratory, UK), unpublished work, 1989.
Mn - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Mn - MnO	P. Chartrand, CRCT, 2021
Mn - P	Bragg-Williams R-K Polynomial	Valid for Mn-Mn ₃ P region	SGTE
Mn - Pb	Bragg-Williams R-K Polynomial		SGTE

Mn - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	Y.-B. Kang, CRCT, 2009
Mn - Sb	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
Mn - Si	MQMPA (Pair Fraction Exp.)	Mn ₅ Si ₂ not present	M.K. Paek, Y-B Kang, CALPHAD 46 (2014), 92-102; A. Shukla, Y.-B. Kang and A.D.
Mn - Sn	Bragg-Williams R-K Polynomial		J. Miettinen: CALPHAD, 2001, 25(1), 43-58
Mn - Ti	Bragg-Williams R-K Polynomial		N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.241-244
Mn - V	Bragg-Williams R-K Polynomial		W. Huang, Met. Trans.A 22A (1991) pp. 1911-1920.
Mn - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Mn - Zn	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
Mn - Zr	Bragg-Williams R-K Polynomial		H.J. Seifert, J. Groebner, F. Aldinger, F.H. Hayes, G. Effenberg, C. Baetzner, H. Flandorfer, P. Rogl, A. Saccone, R. Ferro, Proc.3 rd . International Magnesium Conference, Ed. G. W. Lorimer, Inst.of Materials, London, 1997, pp. 257-270.
Mo - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2003
Mo - Nb	Bragg-Williams R-K Polynomial		COST-507
Mo - Ni	Bragg-Williams R-K Polynomial		COST-507
Mo - P	Bragg-Williams R-K Polynomial		SGTE
Mo - Pt	Bragg-Williams R-K Polynomial		Ph.D. Thesis Tewfik BENLAHARCHE from U. Lorraine (France) 2008
Mo - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2021
Mo - Si	Bragg-Williams R-K Polynomial		P.Y. Chevalier, 2003 Report to SGTE
Mo - Ti	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
Mo - V	Bragg-Williams R-K Polynomial		SGTE
Mo - W	MQMPA (Pair Fraction Exp.)	Ideal liquid	SGTE
Mo - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Mo - Zr	Bragg-Williams R-K Polynomial	Stoichiometric Mo ₂ Zr C15	Perez, B. Sundman, CALPHAD 27 2003, 253-262
Na - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
Na - Ni	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
Na - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Na - Na ₂ O	P. Chartrand, CRCT, 2021
Na - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Na-Na ₂ S.	P. Chartrand, CRCT, 2022
Na - Si	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006 (VLAB Project)
Na - Sn	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", Journal of Chemical Thermodynamics, 66 (2013) 22-33
Na - Ti	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
Na - V	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
Na - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Na - Zn	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", Journal of Chemical Thermodynamics, 66 (2013) 22-33

Na - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Nb - Nd	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Nb - Ni	Bragg-Williams R-K Polynomial		J.-M. Joubert, N.Dupin, B.Sundman CALPHAD 2004
Nb - Si	Bragg-Williams R-K Polynomial		SGTE
Nb - Sn	Bragg-Williams R-K Polynomial		C. Toffolon, Gachon, B. Sundman, JPE 2002
Nb - Ti	Bragg-Williams R-K Polynomial		COST 507, pp.256-260.
Nb - V	Bragg-Williams R-K Polynomial		K.C.H. Kumar, P. Wollants, L. Delaey, Calphad 18 (1994), pp. 71-79.
Nb - W	Bragg-Williams R-K Polynomial		P. Gustafson, Z. Metallkde 79 (1988), pp. 388-396.
Nb - Zn	Bragg-Williams R-K Polynomial		Z. Long, JPED 2016
Nb - Zr	Bragg-Williams R-K Polynomial		A.F. Guillermet, Z. Metallkde. 82 (1991), pp. 478-487.
Ni - O	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Ni - NiO	P. Chartrand, CRCT, 2021
Ni - P	Bragg-Williams R-K Polynomial	Ni-Ni ₆ P ₅	NPL, unpublished work, 1989
Ni - Pb	Bragg-Williams R-K Polynomial		Cui Ping Wang, Xing Jun Liu, I. Ohnuma, R. Kainuma, K. Ishida: CALPHAD, 2000, 24(2), 149-167
Ni - Pd	Bragg-Williams R-K Polynomial		COST-531
Ni - Pt	Bragg-Williams R-K Polynomial		X.-G. Lu et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 33 (2009) 450-456
Ni - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P.Waldner and A.D.Pelton, Z. fur Metallk. 95[8] (2004), 672-681
Ni - Sb	Bragg-Williams R-K Polynomial		Y. Zhang et al., CALPHAD 32 (2008) 378–388
Ni - Si	Bragg-Williams R-K Polynomial	Spurious Ni ₃ Si L1 ₂ below 120°C	M. Lindholm, B. Sundman, Met.Trans.A 26A (1996), pp. 2897-2903.
Ni - Sn	Bragg-Williams R-K Polynomial	Stoichiometric α-Ni ₃ Sn ₂ (lt)	COST-531
Ni - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Ni - Ti	Bragg-Williams R-K Polynomial		C.S. Oh, J. Korean Inst.Met.Mater. 33 (1995), pp. 129-136.
Ni - V	Bragg-Williams R-K Polynomial		COST 507, pp. 261-263.
Ni - W	Bragg-Williams R-K Polynomial		SGTE
Ni - Zn	Bragg-Williams R-K Polynomial		SGTE
Ni - Zr	Bragg-Williams R-K Polynomial		G. Ghosh, J.Mater.Res. 9 (1994), pp. 598-616.
O - Pb	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database Valid for Pb - PbO	P. Chartrand, CRCT, 2021
O - Si	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database Valid for Si - SiO	P. Chartrand, CRCT, 2021
O - Ti	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database Valid for Ti - Ti ₂ O	P. Chartrand, CRCT, 2021
O - V	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database Valid for V - V ₂ O	P. Chartrand, CRCT, 2021
O - Zn	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database Valid for Zn - ZnO	P. Chartrand, CRCT, 2021
P - Sb	Bragg-Williams R-K Polynomial		I.Ansera, C.Chatillon, Calphad 18 (1994) 208
P - Si	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
P - Sn	Bragg-Williams R-K Polynomial		J..Miettinen, Calphad, Vol. 25, No. 1, pp. 67-78
P - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Pb - Pd	Bragg-Williams R-K Polynomial		COST-531

Pb - Pt	Bragg-Williams R-K Polynomial		Z.H.Long et al., JPEDAV (2009) 30:318–322
Pb - S	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Pb - Sb	Bragg-Williams R-K Polynomial		H.Ohtani, K.Okuda, K.Ishida, J.Phase Equilibria 16 (1995) 416-429
Pb - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Pb - Si	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022; based on R.W.Olesinski, G.J.Abbaschian, - TIBull.Alloy Phase Diags. 5 (1984) 271-273
Pb - Sn	Bragg-Williams R-K Polynomial		Based on H.Ohtani, K.Okuda, K.Ishida, J.Phase Equilib.16 (1995) 416-429
Pb - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Pb - Ti	Bragg-Williams R-K Polynomial		unpublished assessment by I. Ansara, H.L. Lukas and S. G. Fries (SGTE)
Pb - Zn	Bragg-Williams R-K Polynomial		T.Jantzen, P.J.Spencer, Calphad 22 (1998) 417-434
Pb - Zr	Bragg-Williams R-K Polynomial		Dixon P. R., Argent B. B., Chart T. G.: CALPHAD, 1998, 22(3), 397-416
Pd - Pt	Bragg-Williams R-K Polynomial		K. Hack, GTT-Technologies, 1995
Pd - Sn	Bragg-Williams R-K Polynomial		P. Chartrand, CRCT, 2023 based on solids from J.Hu et al., Mater. Res. Express 9 (2022) 016507
Pd - Ti	Bragg-Williams R-K Polynomial		P. Chartrand 2018 (Liq + BCC + HCP + FCC) K.Hack, GTT-Technologies, 1996;
Pt - Si	Bragg-Williams R-K Polynomial		L.L. Xu et al. / Computer Coupling of Phase Diagrams and Thermochemistry 32 (2008) 101–105
Pt - Sn	MQMPA (Pair Fraction Exp.)		COST-531
Pt - Ti	Bragg-Williams R-K Polynomial		K. Hack, GTT-Technologies, 1996
Pt - V	Bragg-Williams R-K Polynomial		C.P. Wang et al. / Intermetallics 16 (2008) 544-549
S - Sb	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
S - Se	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2021
S - Si	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2021
S - Sn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
S - Te	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
S - Zn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database	P. Chartrand, CRCT, 2018
Sb - Se	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2021
Sb - Si	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
Sb - Sn	Bragg-Williams R-K Polynomial		C.S.Oh, J.H.Shim, B.J.Lee, D.N.Lee, J.Alloys and Cpd. 238 (1996) 155-166
Sb - Zn	Bragg-Williams R-K Polynomial		L.A.Zabdyr, Calphad 21 (1997) 349-358
Se - Te	Bragg-Williams R-K Polynomial		G. Ghosh, R. C. Sharma, D. T. Li, Y. A. Chang: J. Phase Equil., 1994, 15(2), 213-224
Si - Sn	Bragg-Williams R-K Polynomial		M.H.G.Jacobs, P.J.Spencer, Calphad 20 (1996) 89-91
Si - Te	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2021
Si - Ti	MQMPA (Pair Fraction Exp.)		FSstel
Si - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Si - V	Bragg-Williams R-K Polynomial		COST 507, pp. 270-273.

Si - W	Bragg-Williams R-K Polynomial		C. Vahlas, P.Y. Chevalier, E. Blanquet, Calphad 13 (1989), pp. 273-292.
Si - Zn	MQMPA (Pair Fraction Exp.)		A. Shukla, Y.-B. Kang and A.D. Pelton, Calphad 32 (2008) pp. 470-477
Si - Zr	Bragg-Williams R-K Polynomial		COST 507, pp. 280-283.
Sn - Te	MQMPA (Pair Fraction Exp.)		M. Paliwal, 2021
Sn - Ti	Bragg-Williams R-K Polynomial		F.Hayes, COST 507 (1998) ISBN 92-828-3902-8, p.284-28
Sn - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Sn - V	MQMPA (Pair Fraction Exp.)		Chen, Gierlotka, and Chen, Journal of ELECTRONIC MATERIALS, Vol. 37, No. 11, 2008
Sn - Zn	MQMPA (Pair Fraction Exp.)		M. Medraj, Concordia, 2011
Sn - Zr	Bragg-Williams R-K Polynomial	Spurious inverted L-L miscibility gap above 1700°C in the Sn-rich region	J.Korb, K.Hack, COST 507 (1998) ISBN 92-828-3902-8, p.290-292
Te - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
Ti - V	MQMPA (Pair Fraction Exp.)		Liquid from Zhijun Zhu, CRCT, 2021; Solids from N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.297-298
Ti - W	Bragg-Williams R-K Polynomial		N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.297-298
Ti - Zn	Bragg-Williams R-K Polynomial		K. Doi, S. Ono, H. Ohtani, M. Hasebe: J. Phase Equilib. Diff., 2006, 27(1), 63-74
Ti - Zr	MQMPA (Pair Fraction Exp.)		Liquid from P. Chartrand, CRCT, 2022; Solids from K.C.H. Kumar, P. Wollants, L. Delaey, J.Alloys and Compounds, 225 (1994), pp.121-127.
Tl - Zn	Bragg-Williams R-K Polynomial		S. S. Kim, T. H. Sanders, Z. Metallkunde 2003
V - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014 changed to MQMPA from FSstel
V - Zn	MQMPA (Pair Fraction Exp.)	estimated	P. Chartrand, 2018 (L. est. From Fe-Zn)
V - Zr	Bragg-Williams R-K Polynomial		COST 507, pp. 303-304.
W - Zr	MQMPA (Pair Fraction Exp.)		Liquid from P. Chartrand, CRCT, 2022; Solids from SGTE
Zn - Zr	Bragg-Williams R-K Polynomial		SGTE